

Determining the photodesorption rate of CO and H₂O ice *-a novel approach-*

Daniel M. Paardekooper

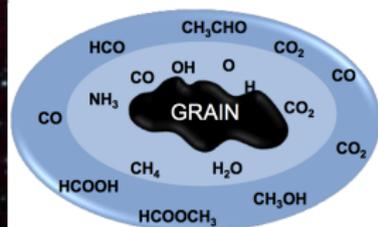
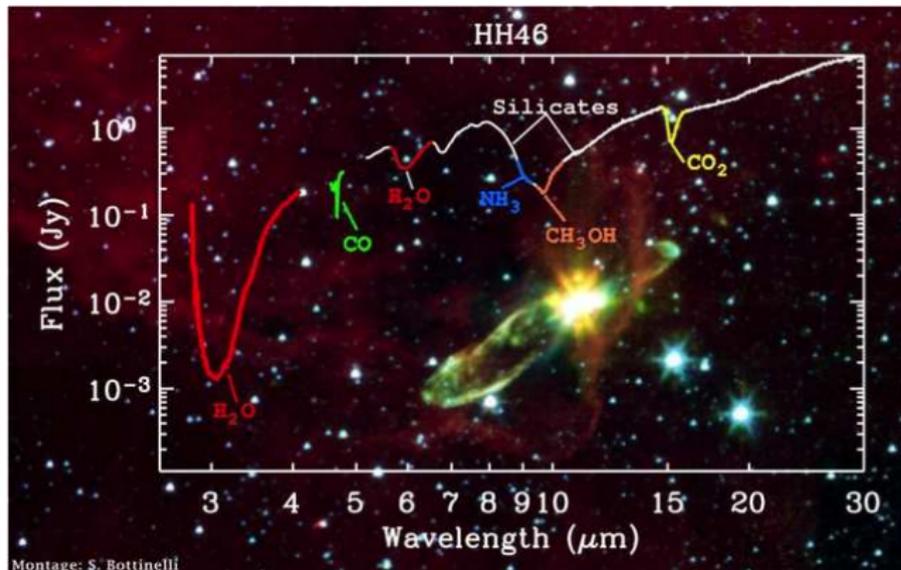
G. Fedoseev, A. Riedo and H. Linnartz

Sackler Laboratory for Astrophysics, Leiden University

dmpaardekooper@strw.leidenuniv.nl

April 12, 2016

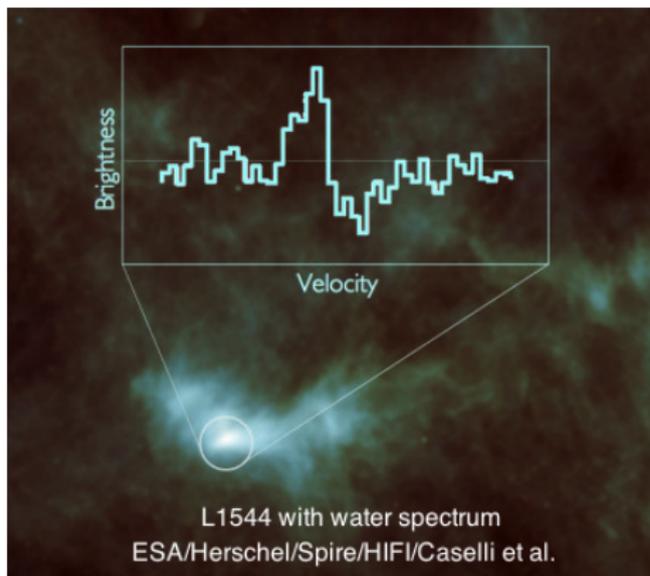
Ices in star forming regions



$T = 10\text{-}20\text{ K}$

- Molecules are found in the gas and solid phase.
- Important species in the solid phase: H_2O , CO , CH_3OH ...

Gas phase species observed in cold regions



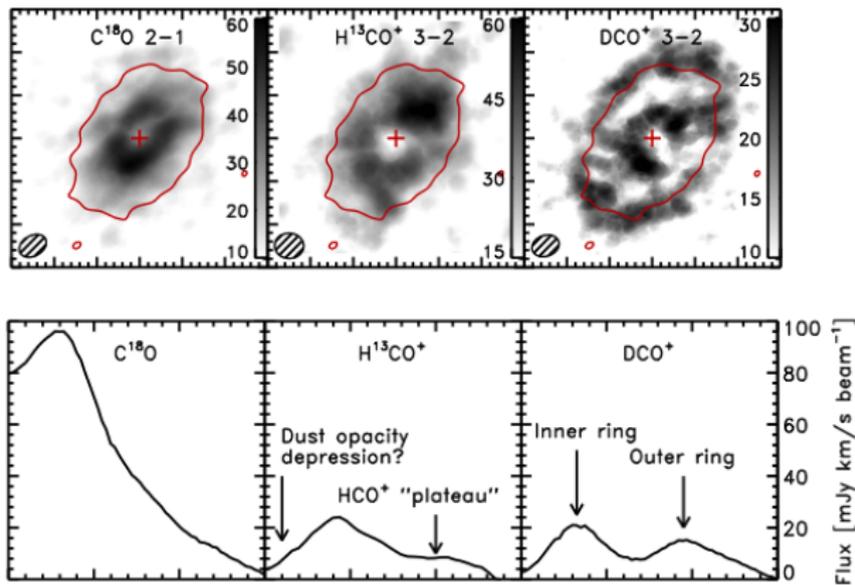
Water in prestellar core

- Gas phase formation route

Non-thermal desorption from dust grains by:

- Cosmic ray spot heating
- Exothermic chemistry
- Electron induced desorption
- VUV induced photodesorption

Gas phase species observed in cold regions

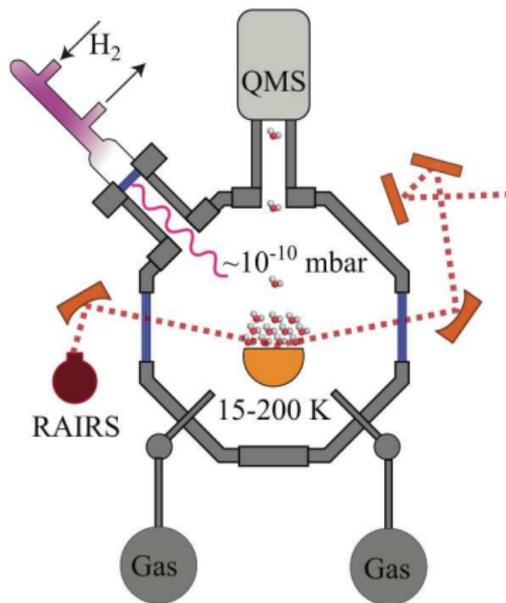


Double ring structure explained by thermal and VUV photodesorption.
Öberg et al. (2015)

Photodesorption in the lab

Broadband discharge lamps (H_2) are used as VUV light source.

Detection scheme: infrared (IR) spectroscopy in conjunction with quadrupole mass spectrometry (QMS).



Cryopad [Öberg et al.]

Literature on photodesorption

CO: broadband discharge lamp (H_2) is used as VUV light source:

	Öberg (2007)	Muñoz Caro (2010)	Chen (2014)
Photodesorption rate (molecules/photon)	$(2.7 \pm 1.3) \times 10^{-3}$	$(3.5 \pm 0.5) \times 10^{-2}$	$(2.12 \pm 0.03) \times 10^{-1}$

CO Photodesorption value spans 3 orders of magnitude

Literature on photodesorption

CO: broadband discharge lamp (H_2) is used as VUV light source:

	Öberg (2007)	Muñoz Caro (2010)	Chen (2014)
Photodesorption rate (molecules/photon)	$(2.7 \pm 1.3) \cdot 10^{-3}$	$(3.5 \pm 0.5) \cdot 10^{-2}$	$(2.12 \pm 0.03) \cdot 10^{-1}$

CO: Synchrotron as photon source, see Fayolle (2011)

8.2 eV $(2.8 \pm 1.7) \cdot 10^{-2}$ molecules/photon

10.2 eV $(6.9 \pm 2.4) \cdot 10^{-3}$ molecules/photon

CO Photodesorption value spans 3 orders of magnitude

Literature on photodesorption

CO: broadband discharge lamp (H_2) is used as VUV light source:

	Öberg (2007)	Muñoz Caro (2010)	Chen (2014)
Photodesorption rate (molecules/photon)	$(2.7 \pm 1.3) \cdot 10^{-3}$	$(3.5 \pm 0.5) \cdot 10^{-2}$	$(2.12 \pm 0.03) \cdot 10^{-1}$

CO: Synchrotron as photon source, see Fayolle (2011)

8.2 eV $(2.8 \pm 1.7) \cdot 10^{-2}$ molecules/photon

10.2 eV $(6.9 \pm 2.4) \cdot 10^{-3}$ molecules/photon

CO Photodesorption value spans 3 orders of magnitude

H_2O photodesorption: $3 \cdot 10^{-3}$ (35 K) Westley et al. (1995)

H_2O photodesorption: $2 \cdot 10^{-3}$ (20 K) Öberg et al. (2010)

Photodesorption in the lab

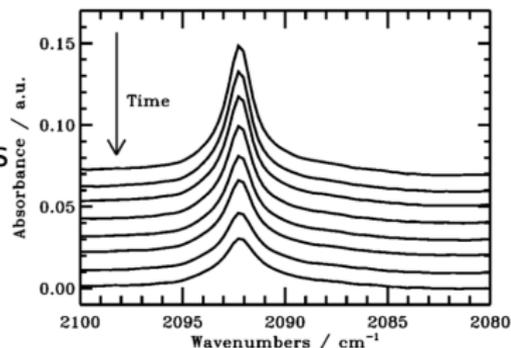
Both methods come with restrictions;

IR spectroscopy:

- Hard to discriminate between signal loss due photodesorption or other processes (e.g. photochemistry)

QMS:

- Challenge to convert the gas-phase abundance into (photo)desorption rate.



Öberg et al. (2007)

Photodesorption in the lab

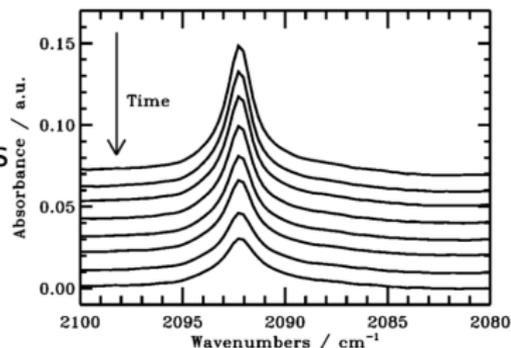
Both methods come with restrictions;

IR spectroscopy:

- Hard to discriminate between signal loss due photodesorption or other processes (e.g. photochemistry)

QMS:

- Challenge to convert the gas-phase abundance into (photo)desorption rate.

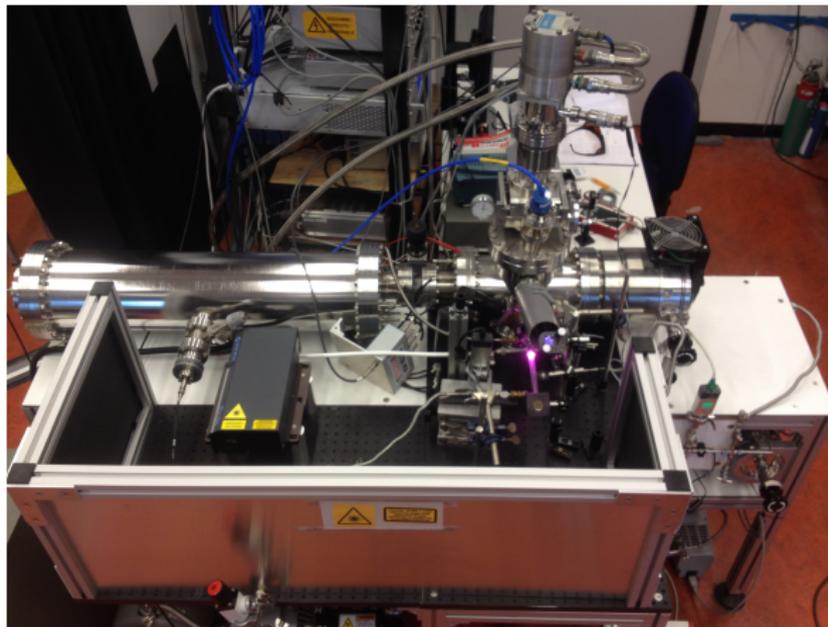


Öberg et al. (2007)

Aim of the present study: Use independent method!

Aim of the present study

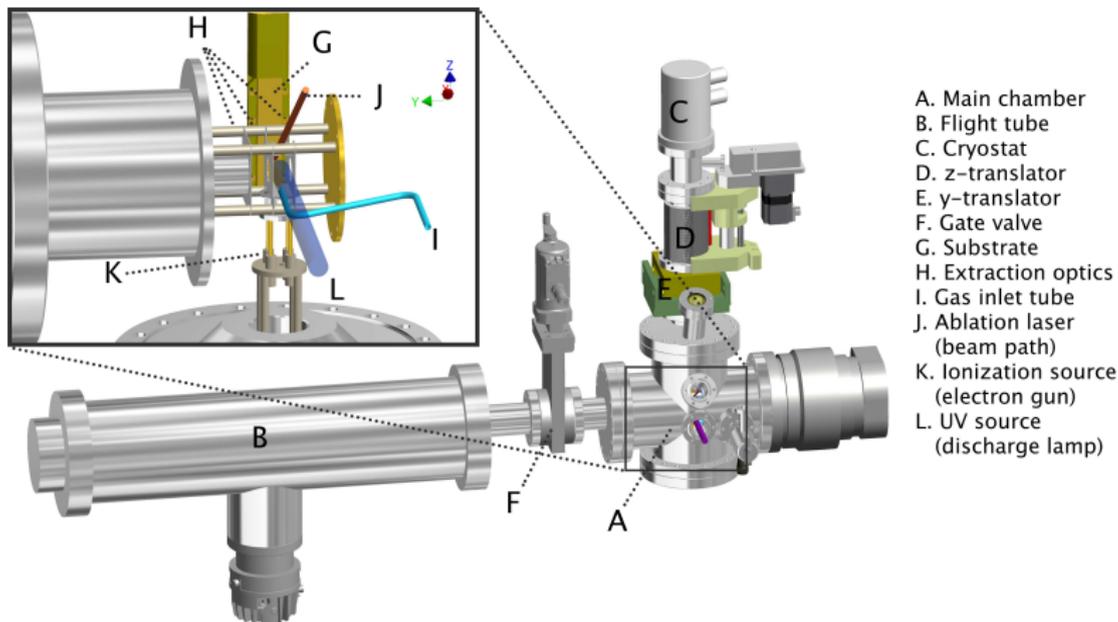
Determine the photodesorption rate with an alternative detection scheme.



MATRI²CES (Mass Analytical Tool for Tracing Reactions in ICES.)
(Paardekooper et al. 2014)

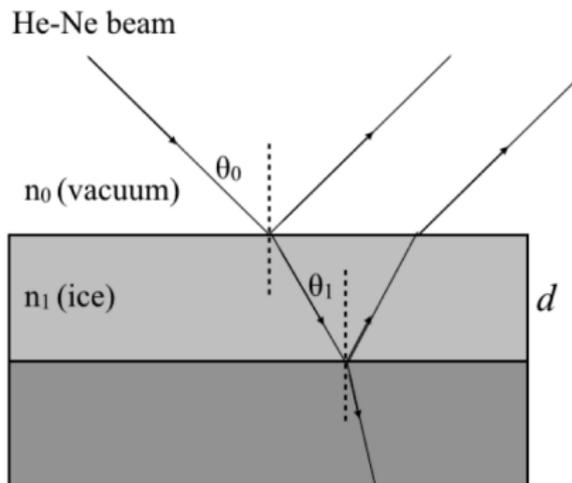
Aim of the present study

Determine the photodesorption rate with an alternative detection scheme.



MATRI²CES (Mass Analytical Tool for Tracing Reactions in ICES.)
(Paardekooper et al. 2014)

Deposition rate calibration (HeNe interference)



$$d = \frac{\lambda}{2n_1/n_0 \cdot \cos(\theta_1)} \quad (1)$$

$$\text{Deposition rate} = \frac{d \cdot \rho \cdot N_a}{M \cdot t} \quad (2)$$

Deposition rate for CO: $1.7 \cdot 10^{13}$ molecules $\text{cm}^{-2} \text{s}^{-1}$

Deposition rate for H₂O: $3.0 \cdot 10^{13}$ molecules $\text{cm}^{-2} \text{s}^{-1}$

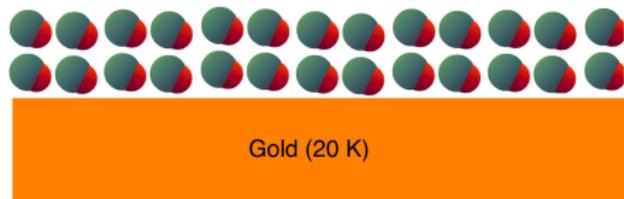
CO at 20 K, $\rho = 0.80 \text{ g cm}^{-3}$ and $n_1 = 1.27$ (Roux et al. 1980)

H₂O at 125 K, $\rho = 0.93 \text{ g cm}^{-3}$ and $n_1 = 1.31$ (Brown et al. 1996)

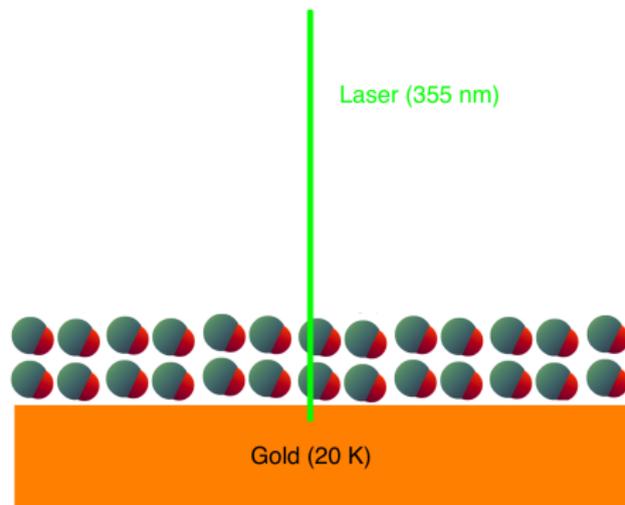
Detection scheme

Gold (20 K)

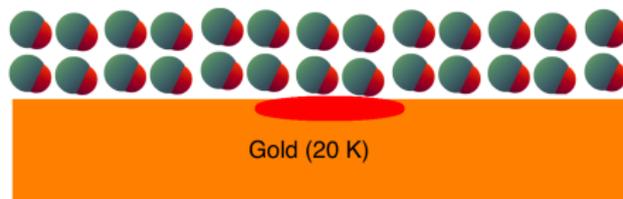
Detection scheme



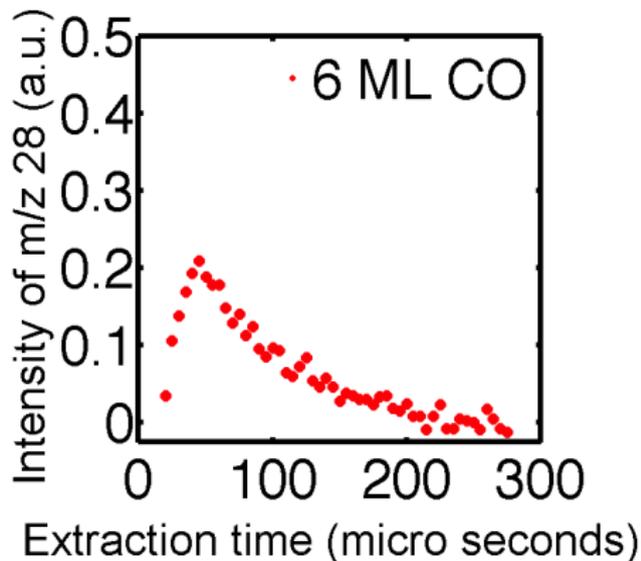
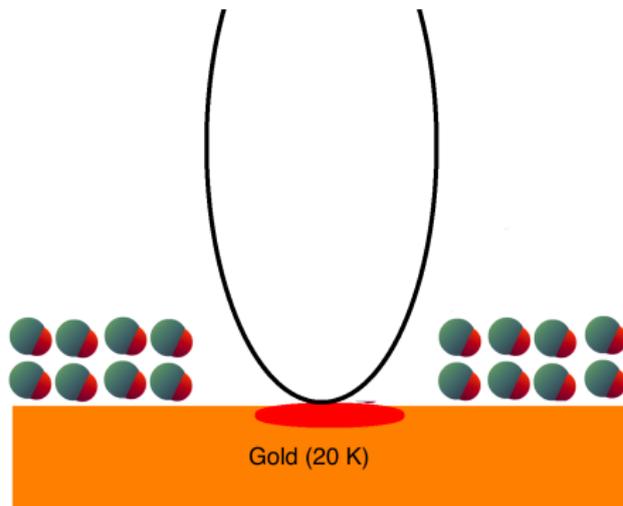
Detection scheme



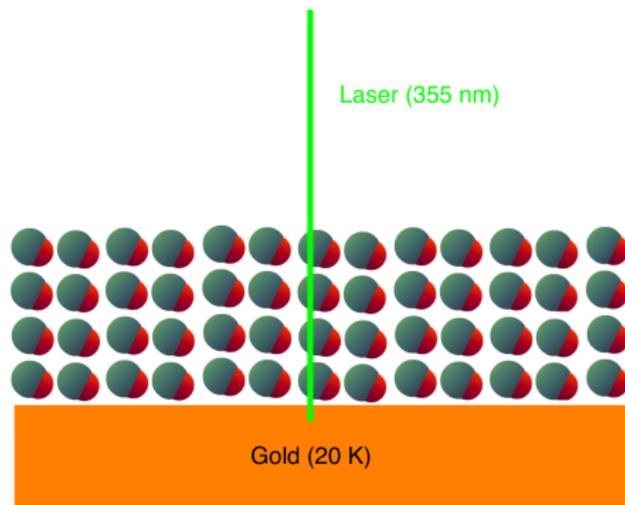
Detection scheme



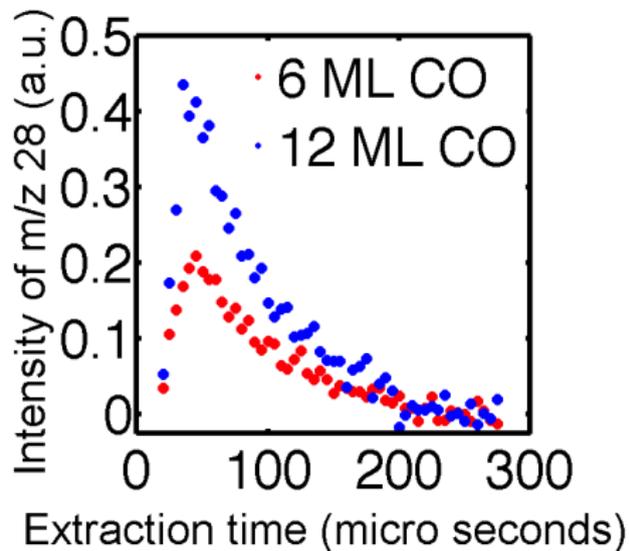
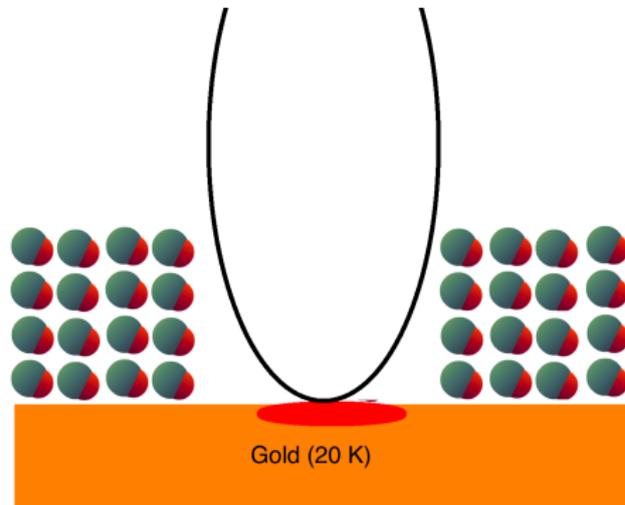
Detection scheme



Detection scheme



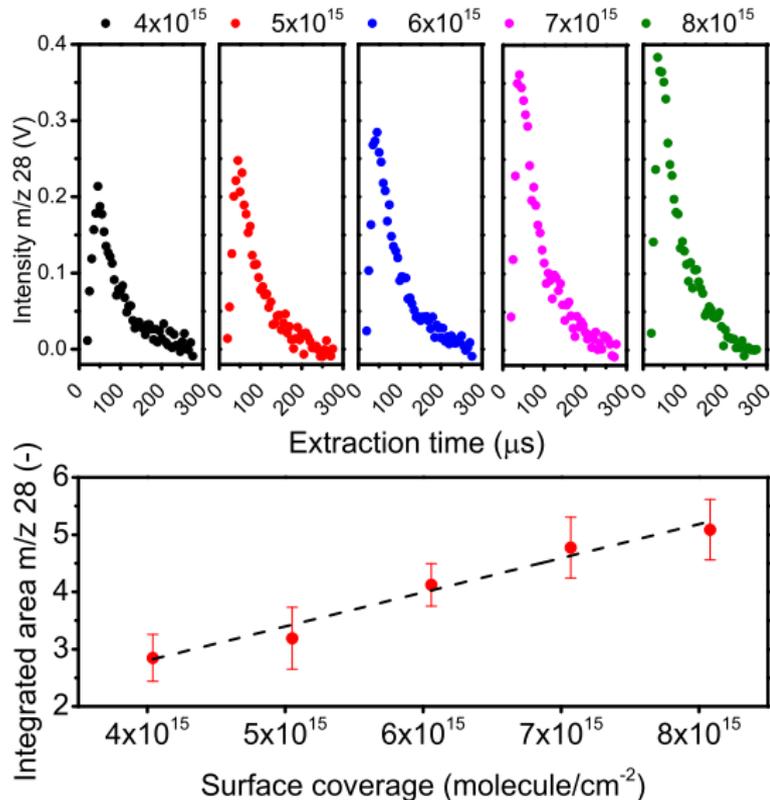
Detection scheme



Plume structure calibration

Different surface coverages show different laser induced desorption plume profiles.

Using this calibration we can trace the surface coverage.

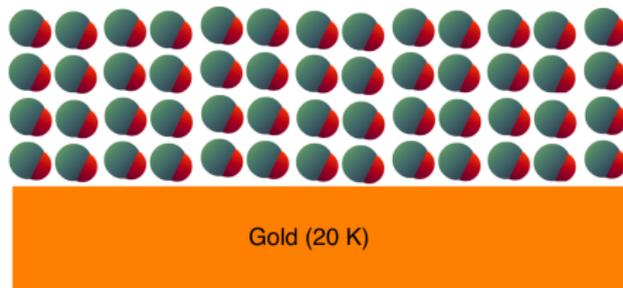


Typical photodesorption experiment

Gold (20 K)

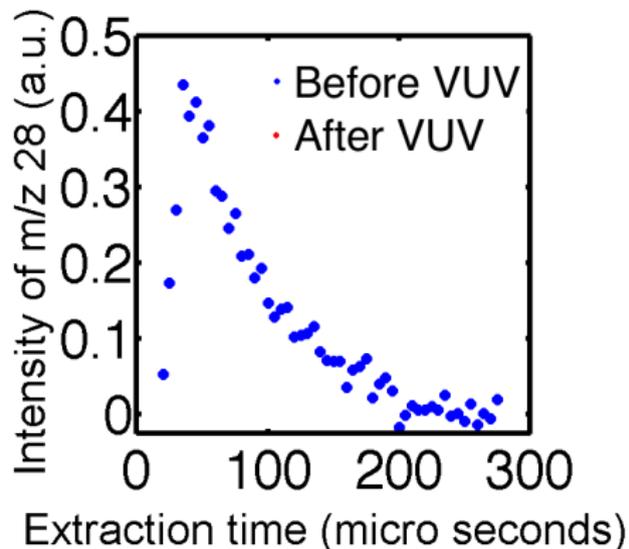
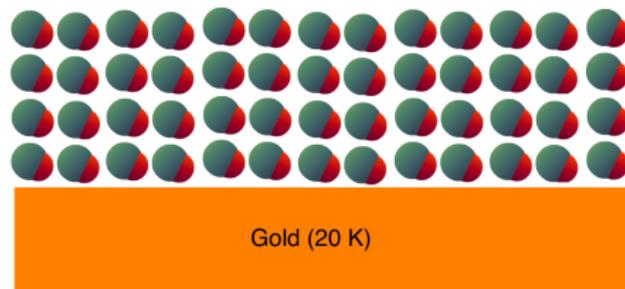
1. Ice deposition with specific surface coverage
- 2.
- 3.

Typical photodesorption experiment



1. Ice deposition with specific surface coverage
- 2.
- 3.

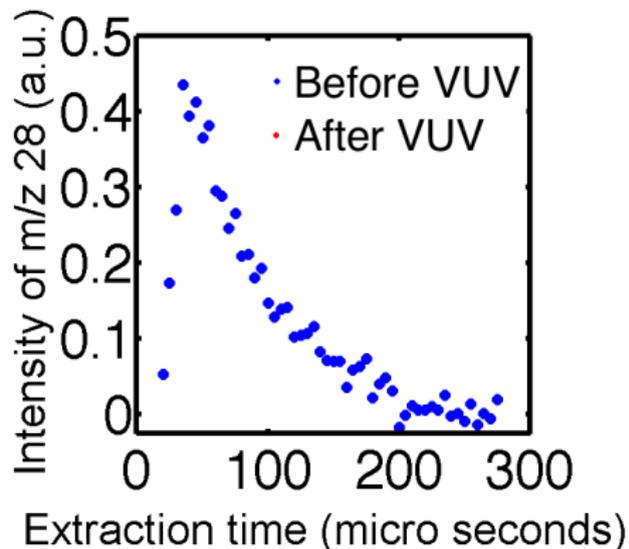
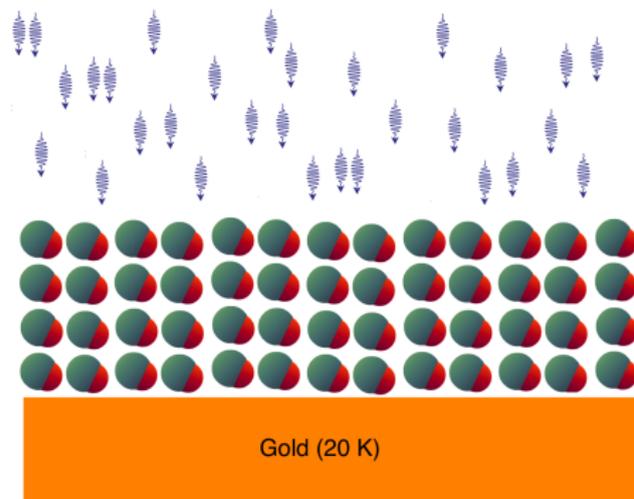
Typical photodesorption experiment



1. Ice deposition with specific surface coverage
2. Obtain spectrum
- 3.

Typical photodesorption experiment

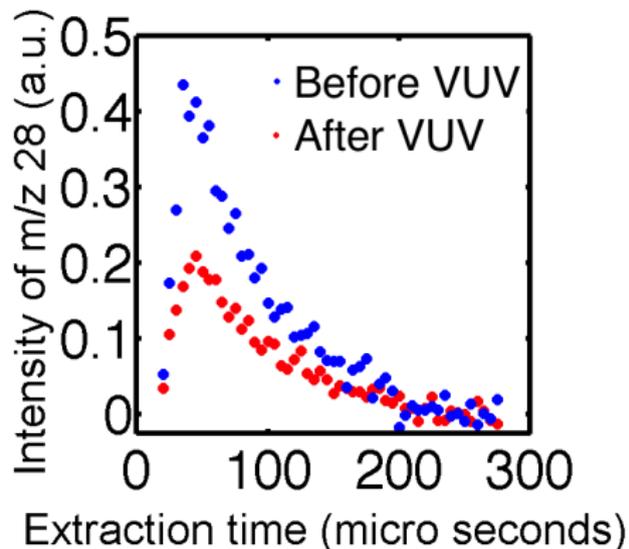
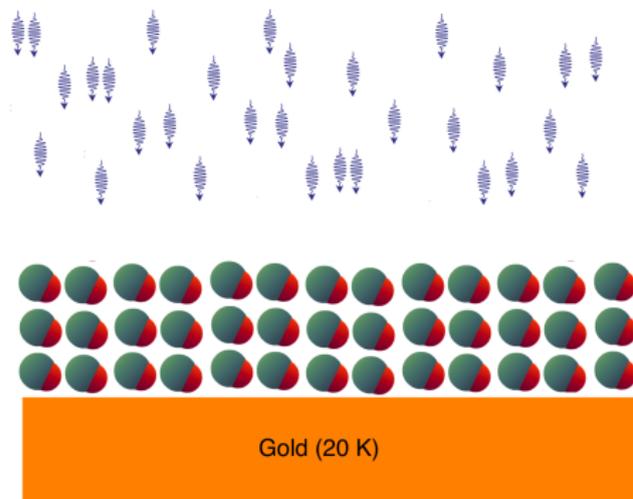
VUV photons from the H₂ lamp



1. Ice deposition with specific surface coverage
2. Obtain spectrum
- 3.

Typical photodesorption experiment

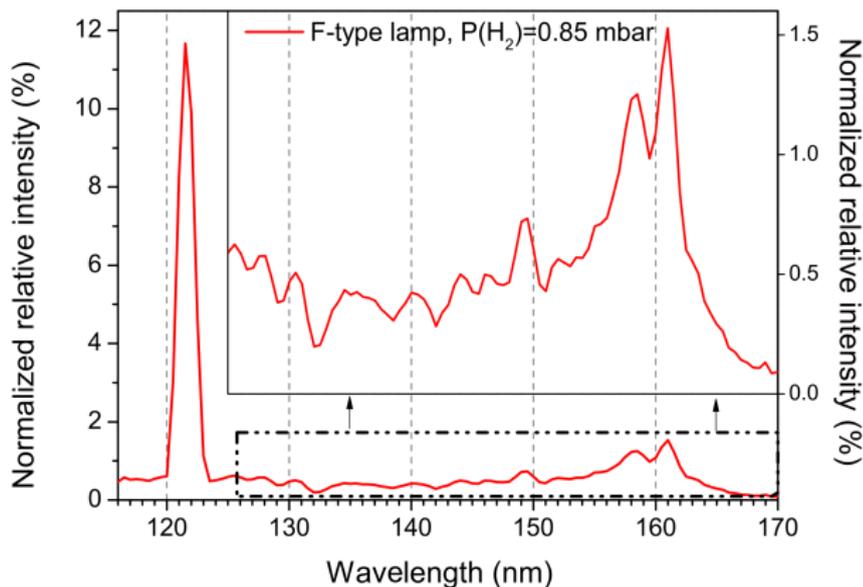
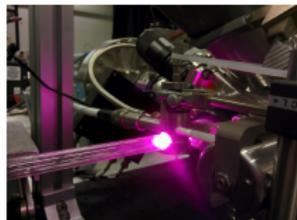
VUV photons from the H₂ lamp



1. Ice deposition with specific surface coverage
2. Obtain spectrum
3. Obtain spectrum after VUV processing

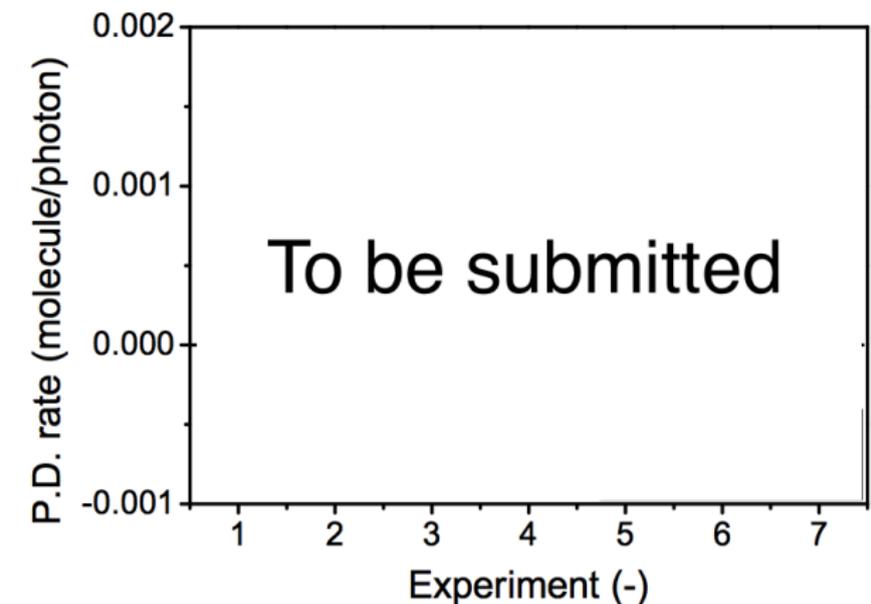
VUV flux calibration (NIST calibrated photodiode)

Absolute spectrum known (Ligterink, Paardekooper, et al. 2015)



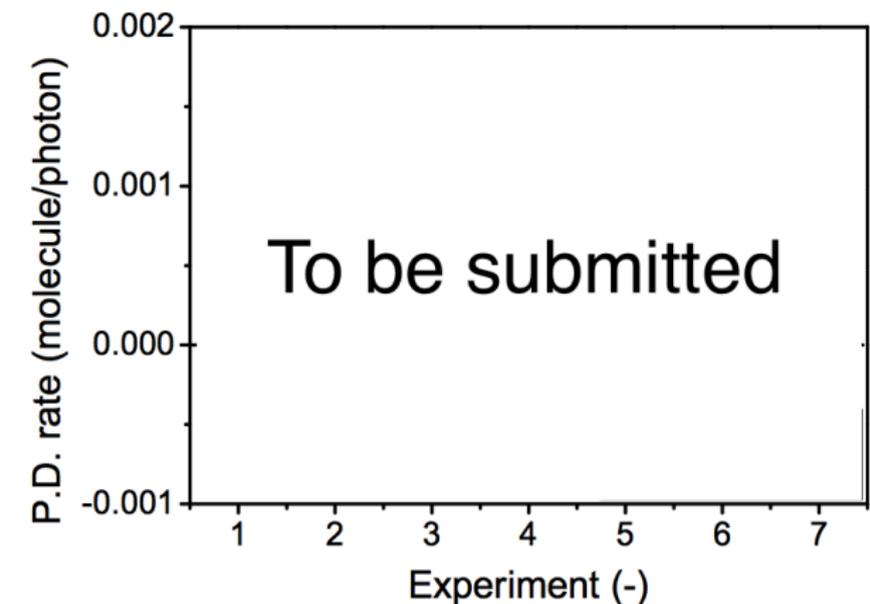
Flux at sample position: $(2.4 \pm 0.7) \cdot 10^{14}$ photon $\text{cm}^{-2}\text{s}^{-1}$

Photodesorption of CO ice (20 K)

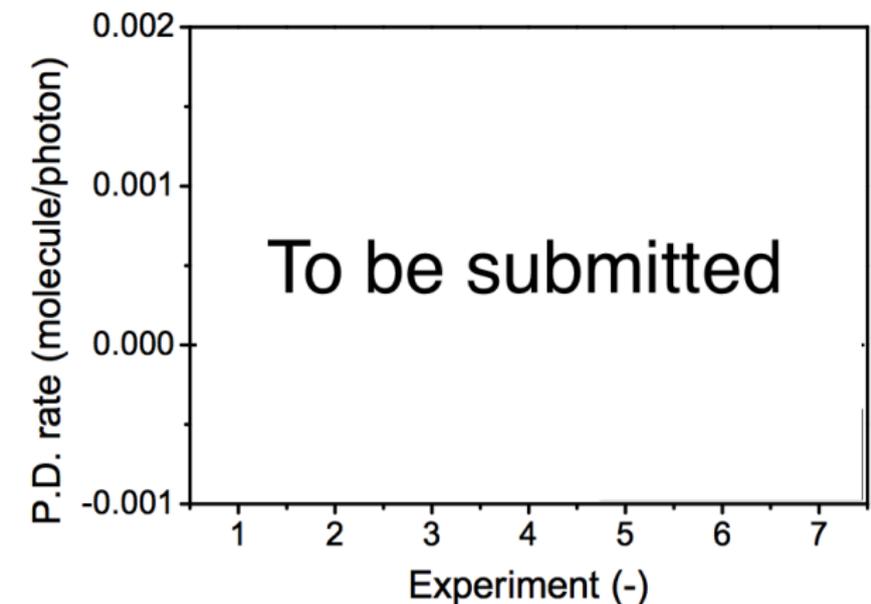


Photodesorption rate [20 K]: $(1.4 \pm 0.7) \cdot 10^{-3}$ molecules/photon
Photochemistry: CO_2 formation $\leq 2\%$ after $5 \cdot 10^{18}$ photons cm^{-2}

Photodesorption of H₂O ice (preliminary results)

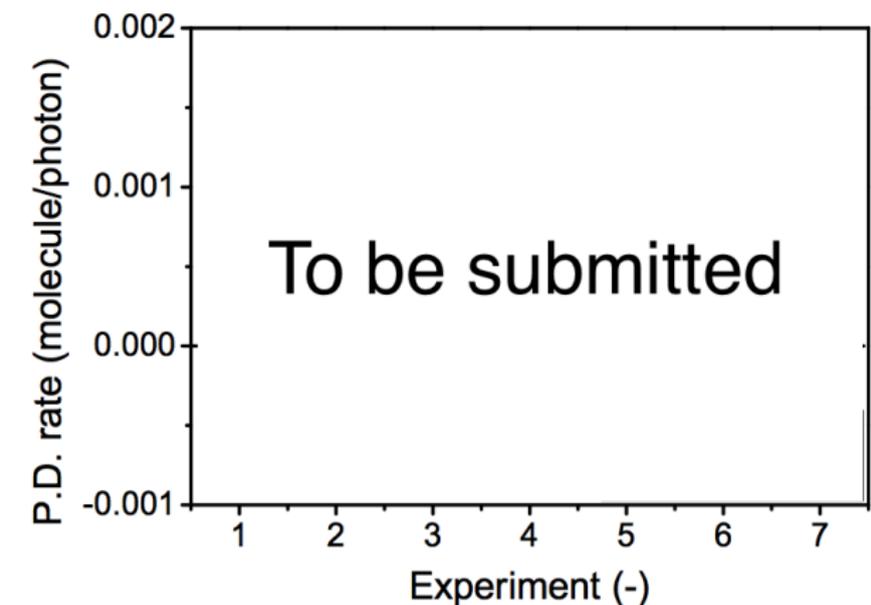


Photodesorption of H₂O ice (preliminary results)



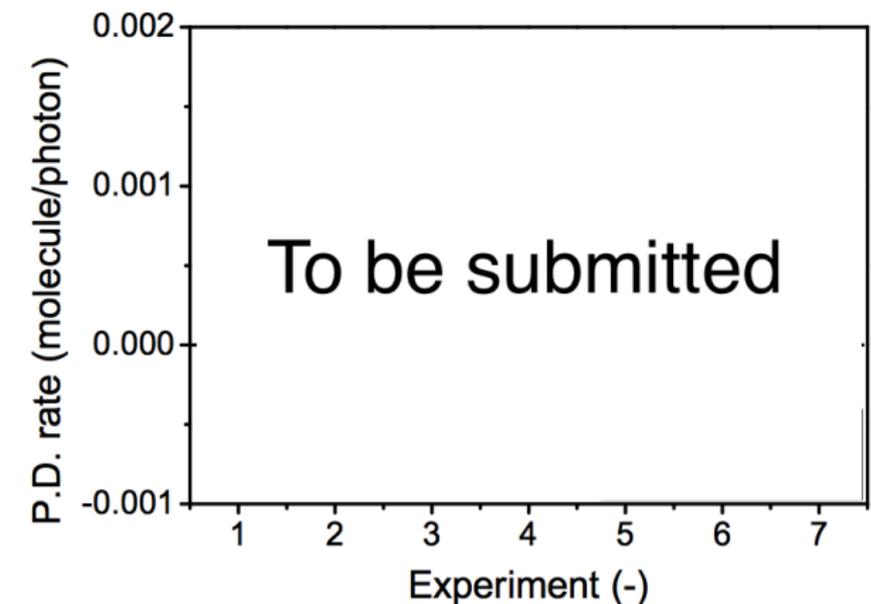
H₂O photodesorption rate [125 K]: $(1.5 \pm 0.7) \cdot 10^{-3}$ molecules/photon

Photodesorption of H₂O ice (preliminary results)



H₂O photodesorption rate [125 K]: $(1.5 \pm 0.7) \cdot 10^{-3}$ molecules/photon

Photodesorption of H₂O ice (preliminary results)



H₂O photodesorption rate [125 K]: $(1.5 \pm 0.7) \cdot 10^{-3}$ molecules/photon

H₂O photodesorption rate [20 K]: $\leq 6 \cdot 10^{-4}$ molecules/photon

MATRI²CES is a, sensitive and versatile tool, the setup can be used as independent method to quantify photodesorption.

- For CO ice at 20 K: the photodesorption rate is $(1.4 \pm 0.7) \cdot 10^{-3}$ molecules/photon.
- CO₂ formation during VUV processing $\leq 2\%$ after $5 \cdot 10^{18}$ photons cm^{-2} .
- For H₂O ice at 125 K: the photodesorption rate is $\sim 1.5 \cdot 10^{-3}$ molecules/photon.
- For H₂O ice at 20 K: we obtained an upper limit of $\sim 6 \cdot 10^{-4}$ molecules/photon, 3 (!) times lower than previously measured by Öberg (2010).

Thank you for your attention

