WATER FORMATION IN INTERSTELLAR ICES

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Study of the different water formation routes through surface chemistry depending on the interstellar conditions

Why dust grains?

- Powerful interstellar catalysts
- Water is one of the most abundant ices on grains
- Adsorption is not enough to explain observations of H₂O ice abundances



Meijerink code

Before

(Meijerink & Spaans 2005)

Chemistry

- Gas chemistry (UMIST 99): 4453 chemical reactions
- Dust chemistry: H₂ formation

Heating & Cooling

- Photo-electric effect on grains
- Carbon ionization
- H2 photo-dissociation
- H2 collisional (de-)excitation
- Gas-grain collisions
- UV & cosmic-ray ionization
- Fine-structure & metastable-lines
- Recombination
- Molecular cooling (H_2, CO, H_2O)

Meijerink code

Before

(Meijerink & Spaans 2005)

Chemistry

- Gas chemistry (UMIST 99): 4453 chemical reactions
- Dust chemistry: H₂ formation



(Esplugues et al. submitted to A&A)

Now

Chemistry

- Gas chemistry (KIDA 2014): 7503 chemical reactions
- Dust chemistry (from laboratory experiments, e.g., Dulieu et al. 2013, Minissale et al. 2015, 2016):
 225 chemical reactions
 - Adsorption
 - Thermal & chemical desorption
 - Two-body reactions
 - Photo-processes
 - Cosmic-ray processes
- 22 solid species:
 H, H₂, O, O₂, OH, O₃, H₂O, HO₂, H₂O₂,
 CO, HCO, H₂CO, H₃CO, H₄CO, CO₂, N,
 N₂, C, CH, CH₂, CH₃, CH₄.

Meijerink code: dust treatment

Icy grains and bare grains (different binding energies)

• Dust temperature (T_d) :

- Impact on gas temperature (T_q)
- Variations in the reaction rates
- Freeze out of gas species
- Ice mantle formation



Garrod & Pauli (2011):

 $T_{dust} = 18.67 - 1.637 [A_v - \log(G_o)] + 0.07518 [A_v - \log(G_o)]^2 - 0.001492 [A_v - \log(G_o)]^3$

Water in different interstellar environments



 G_0 =1 corresponds to a flux of 1.6·10⁻³ erg cm⁻² s⁻¹







Dust-phase: H₂O ice monolayer formation



• High *n* and, especially, low G_0 favour the formation of H_2O ice full monolayers

Water formation: Icy vs Bare grains



- Solid H₂O: small differences between icy and bare grains.
- Gas H₂O: abundances ~1 order of magnitude higher with bare grains.
- Bare grains also favour the formation of solid and gas H_2O_2 .





Conclusions

Chemical desorption and icy-bare dust (different binding energies)

Type of grain substrate:

- Solid water: small abundance differences between bare and icy grains
- Gas water: bare grains favour its formation

Gas water formation precursors (at $A_v \le 5$ mag):

- Solid H and solid OH

→ Surface processes forming gas water:

- PDRs: chemical desorption (A_v≤5 mag)
- Dark clouds: chemical desorption ($A_v < 1.5 \text{ mag}$) & photo-desorption ($1.5 \le A_v \le 5 \text{ mag}$)