# Astrobiology An Overview

Markus Kissler-Patig



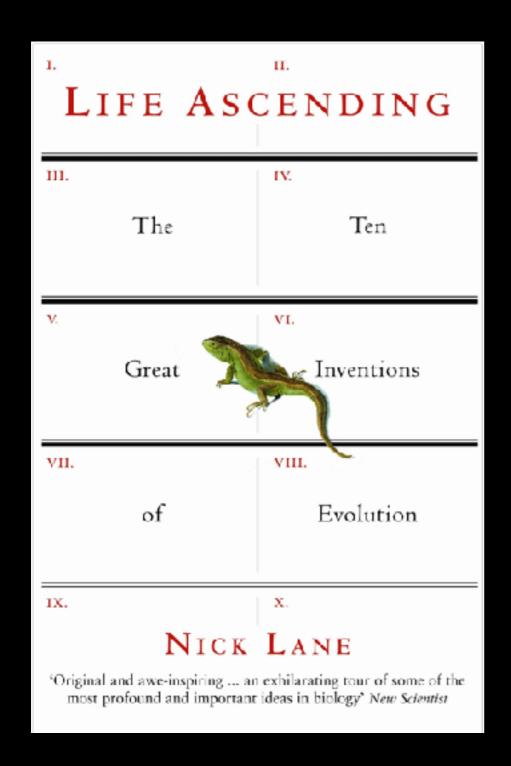
November 20-24, 2023

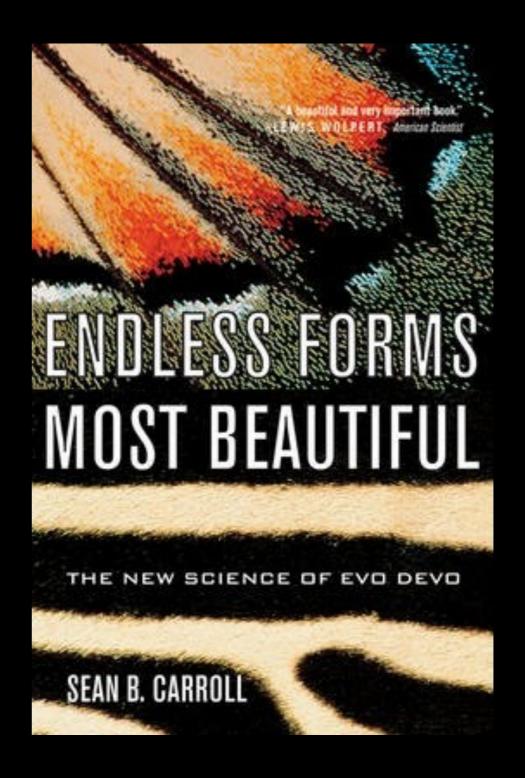
Daily: 10:00-12:00 & 13:00-14:00

https://www.cosmos.esa.int/web/astrobio/imprs-2023



#### Books mentioned yesterday:





# Astrobiology An Overview

Day 2

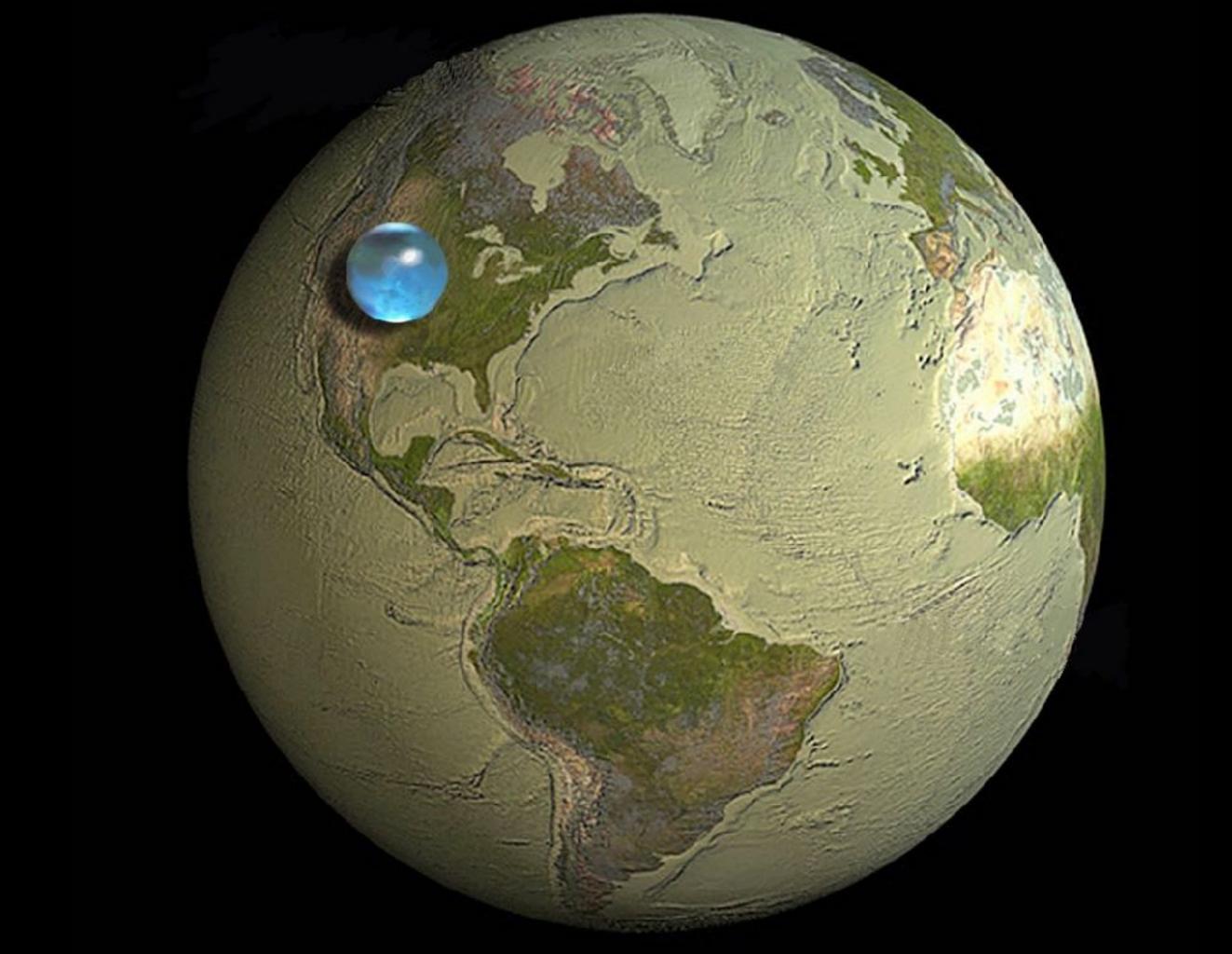
Earth Climate History; Mars and Venus Climates

https://www.cosmos.esa.int/web/astrobio/imprs-2023

Monday November 20	Day 1: Definition of Life; Origin of Life; Evolution of Life; Limits of Life 10:00-12:00 & 13:00-14:00
Tuesday November 21	Day 2: Earth Climate History; Mars and Venus Climates 10:00-12:00 & 13:00-14:00 OLD SEMINAR ROOM
J	Day 3: Habitable Places in the Solar System; Mars; Moons of Giant Planets 10:00-12:00 & 13:00-14:00
	Day 4: Habitable Places beyond the Solar System; Exoplanets properties; Biosignatures 10:00-12:00 & 13:00-14:00
Friday November 24	Day 5: Search for Extraterrestrial Intelligence; Alien Biochemistry 10:00-12:00 & 13:00-14:00

# Basics of Climate





#### If Earth were a black body heated by the Sun:

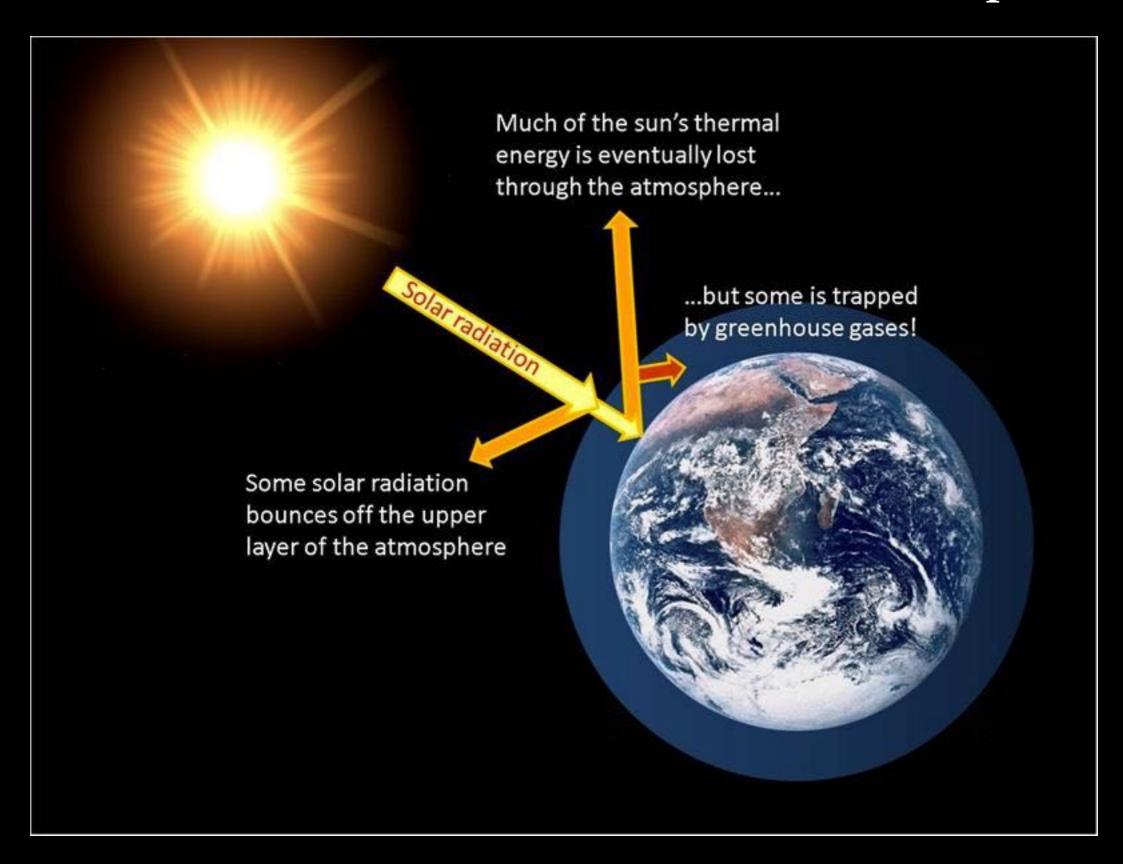
$$T_{\rm eff} = 255 K$$

assuming 0.3 Albedo

But

T<sub>surf</sub> ~ 288K

#### Two heat sources: the sun + the heated atmosphere



# Which is the most important Greenhouse gas for Earth's climate?

#### If Earth were a black body heated by the Sun:

$$T_{eff} = 255K$$

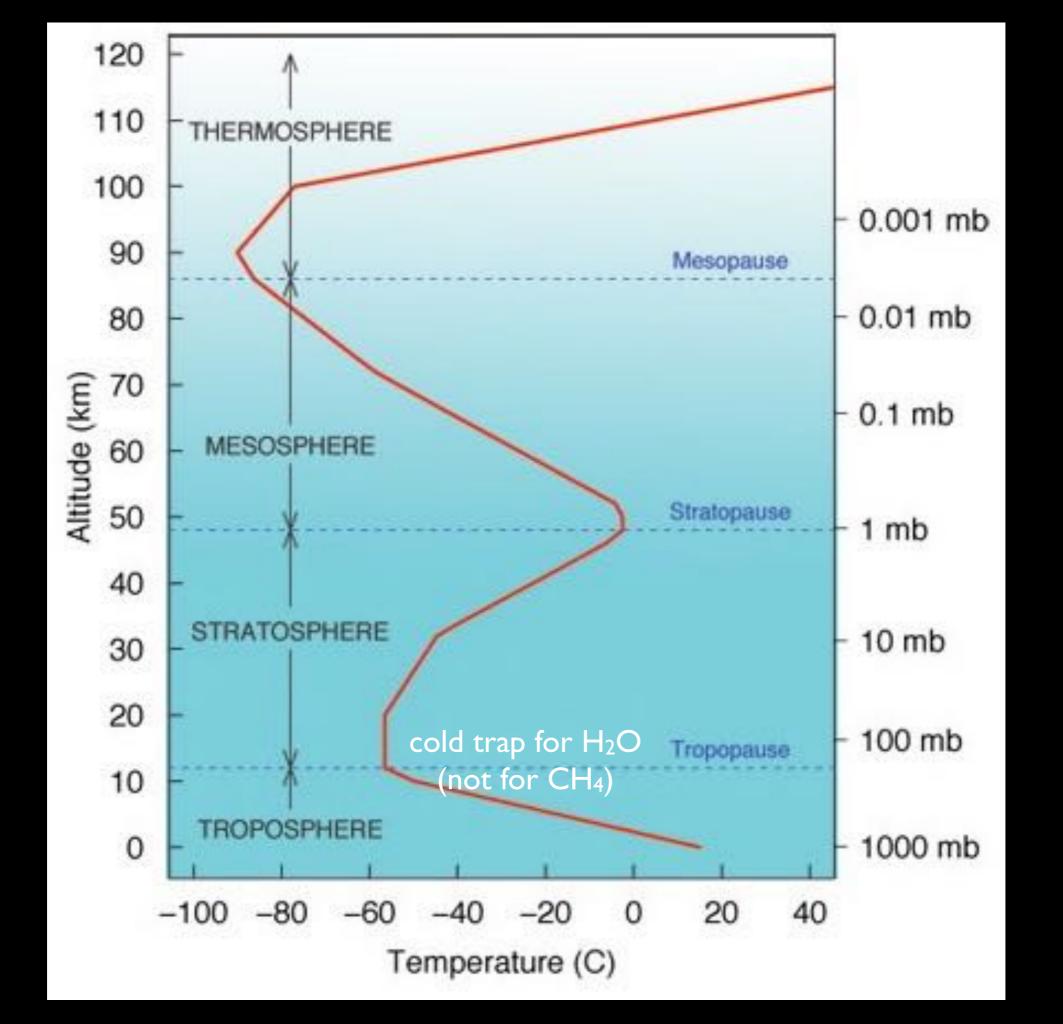
 $T_{\text{eff}} = 255K$  assuming 0.3 Albedo

But

 $T_{\text{surf}} \sim 288K$ 

The Greenhouse effect contributes ~33K  $(20K H_2O, 10K CO_2, 2-3K CH_4, N_2O, O_3, CFCs)$  The Temperature of the planet depends (to first order) on only three factors:

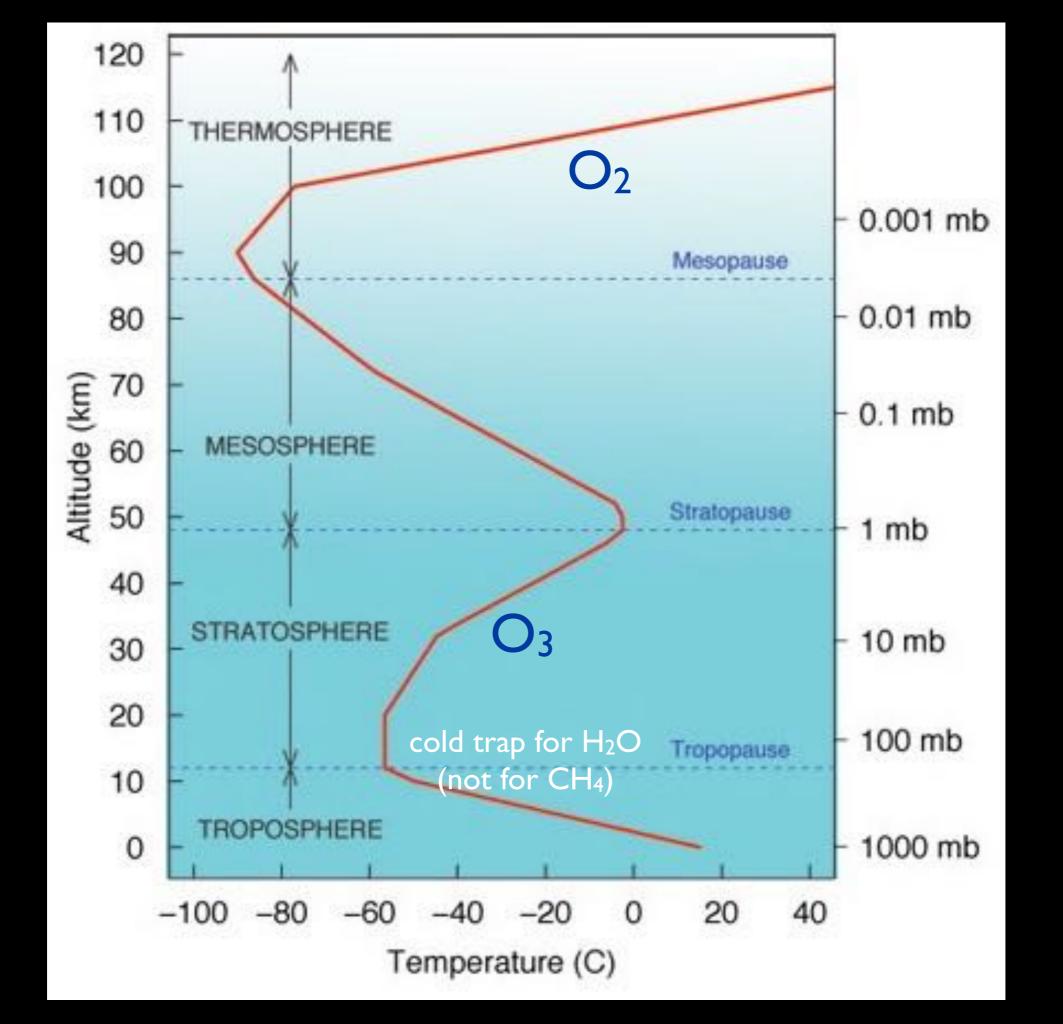
- The Solar flux (geometry, solar physics)
- The Albedo (80% due to clouds, very difficult to model!)
- The Greenhouse effect (atmospheric composition)



#### Grimsvötn: ashes 8-12 km high, H<sub>2</sub>O up to 20km

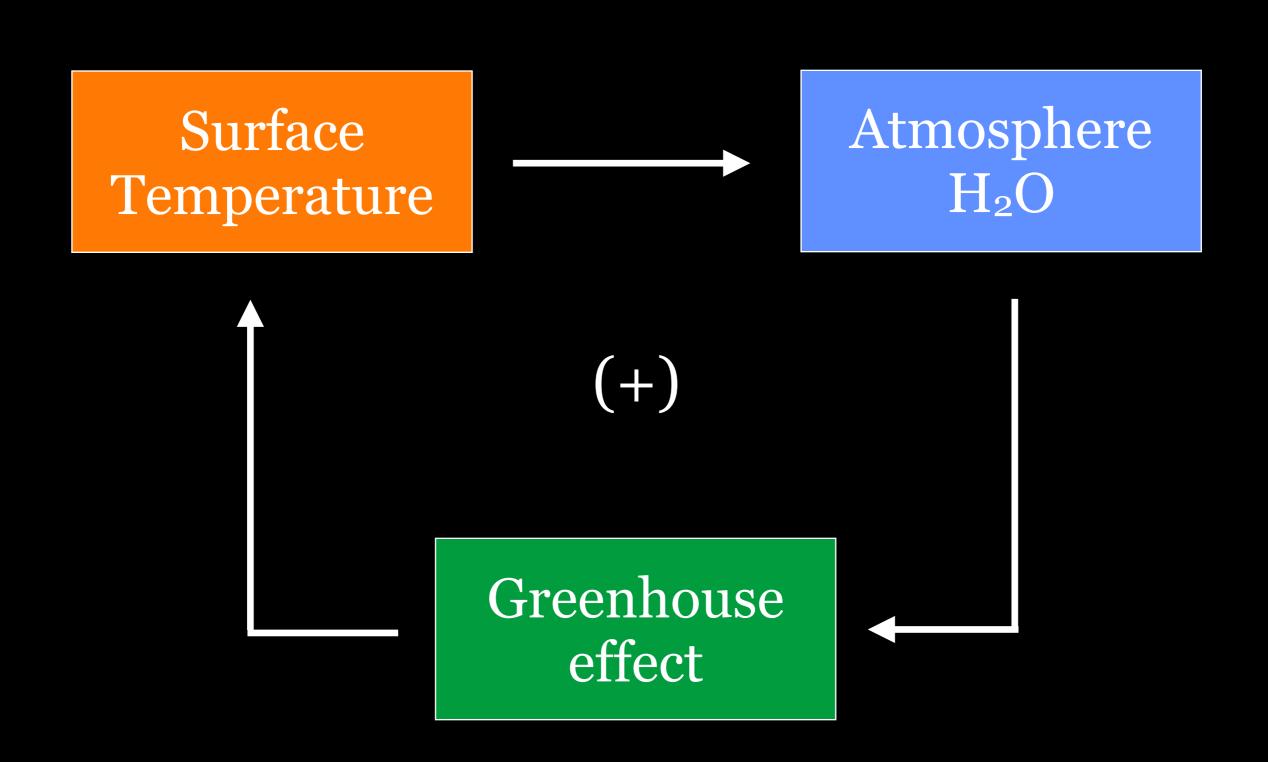


# At which altitude is the Ozone layer?

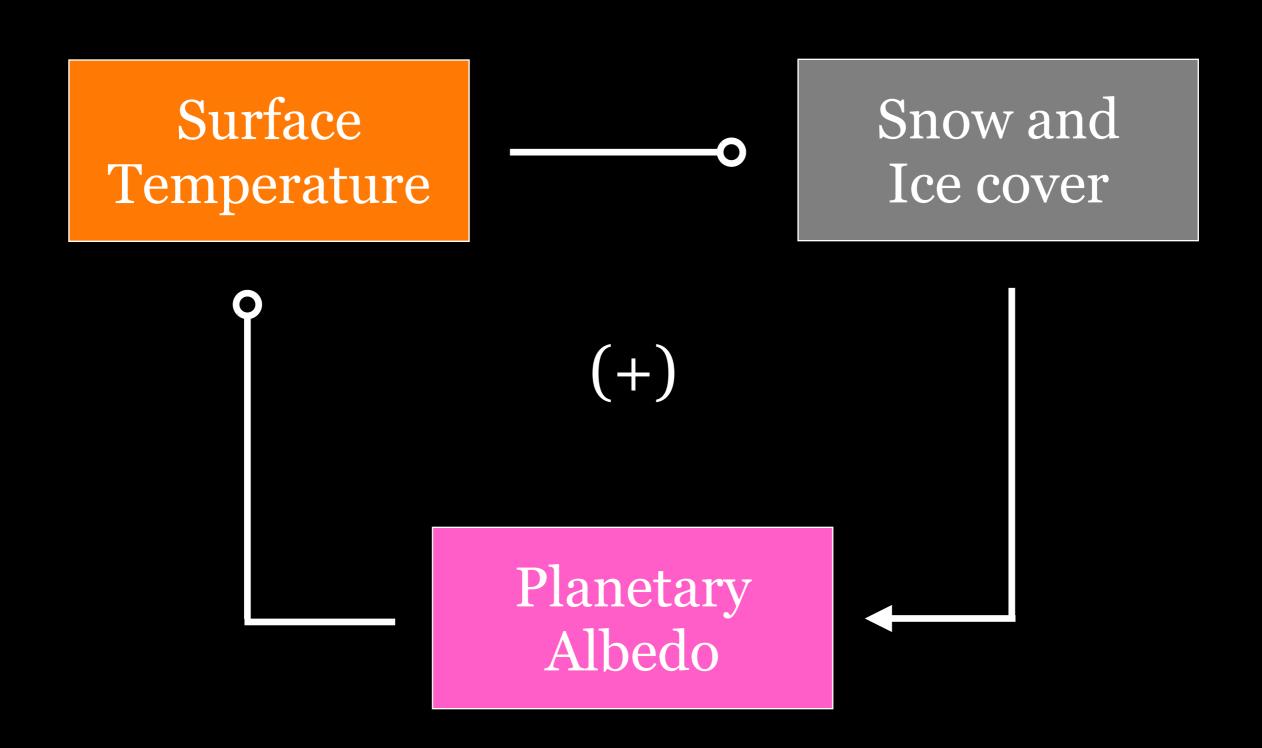


#### Climate Feedback loops

#### Solar flux variations Greenhouse variations



#### Albedo variations



Gained in importance in the last 2 Myr

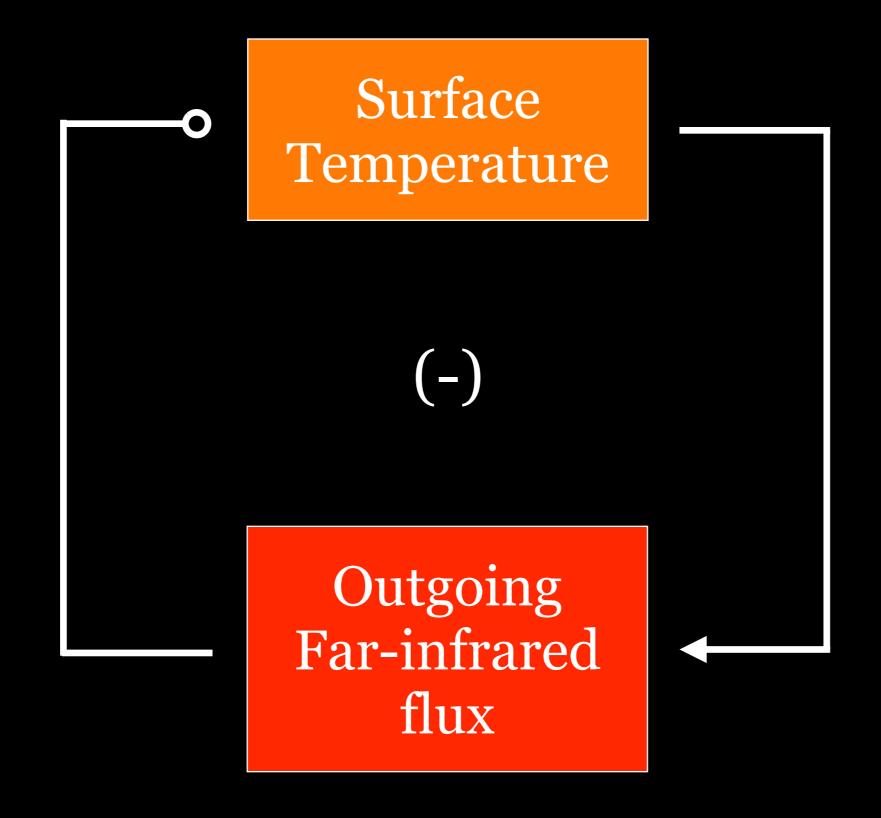
#### POSITIVE (+) Feedback loops are unstable

But the climate on Earth is stable...



⇒ There must be NEGATIVE (-) feedback loops that stabilize the climate (at least on short time scales)

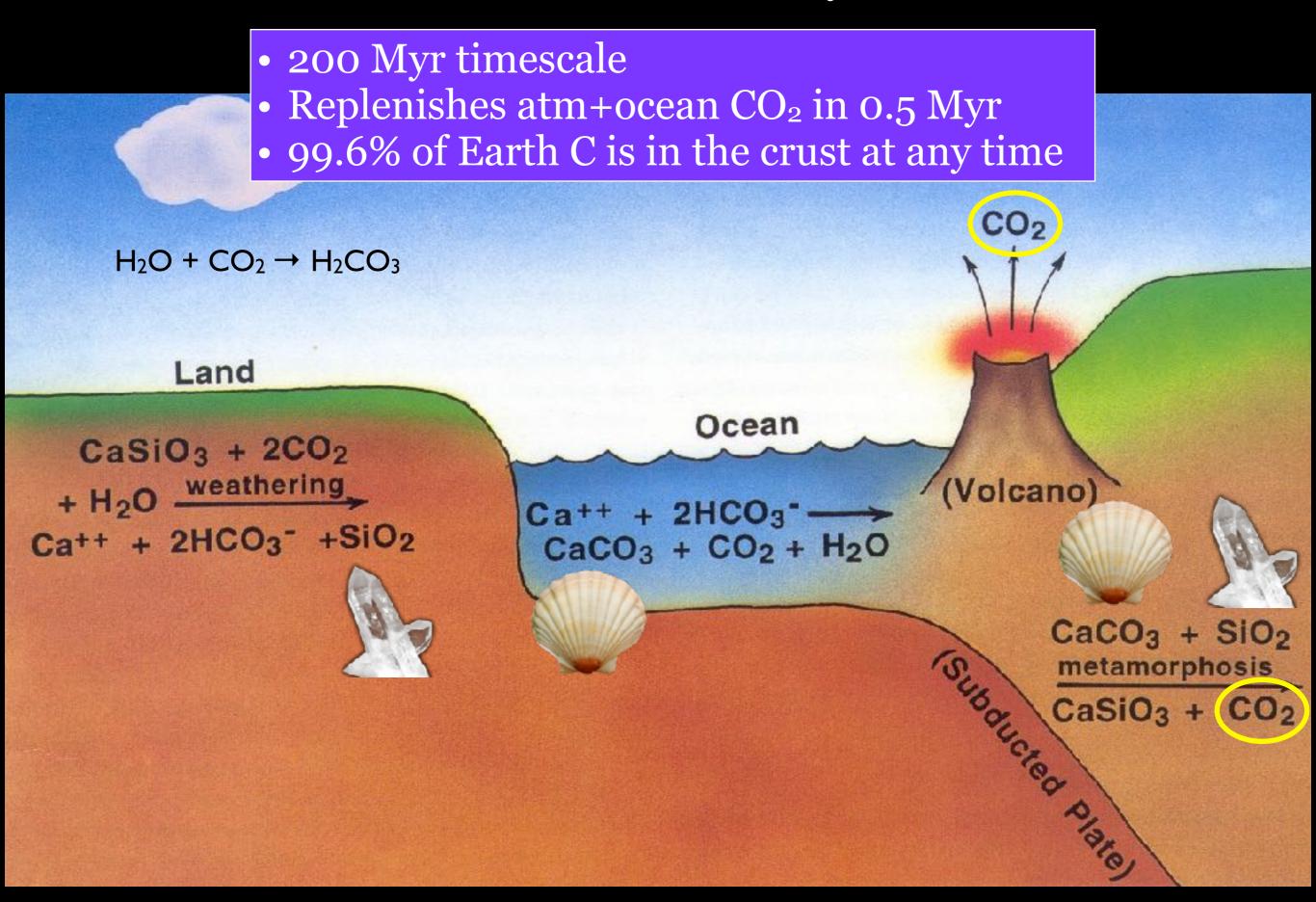
#### Provides short-term stability

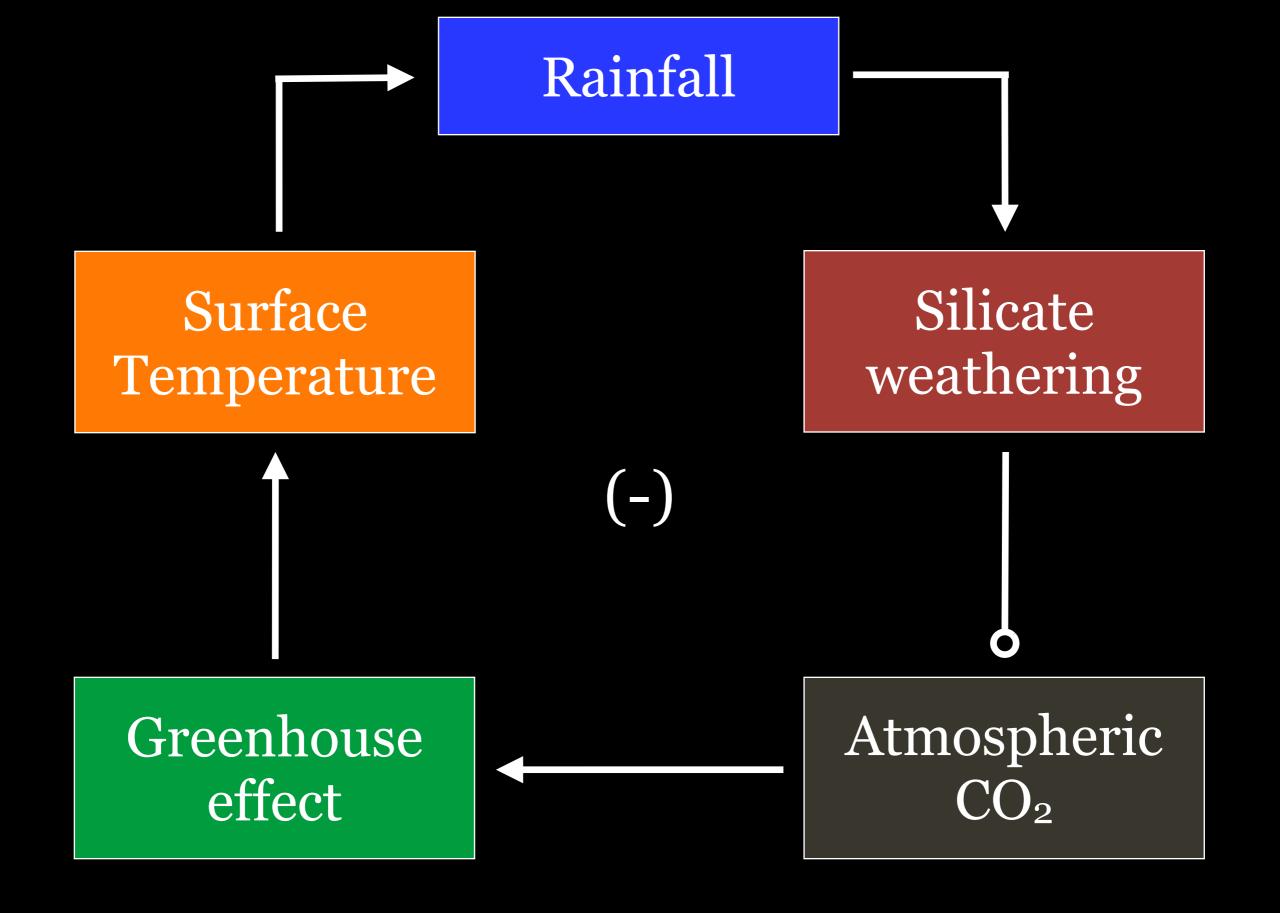


# 'Faint Young Sun' Problem

Do we have evidence for a long-term stabilizing process?

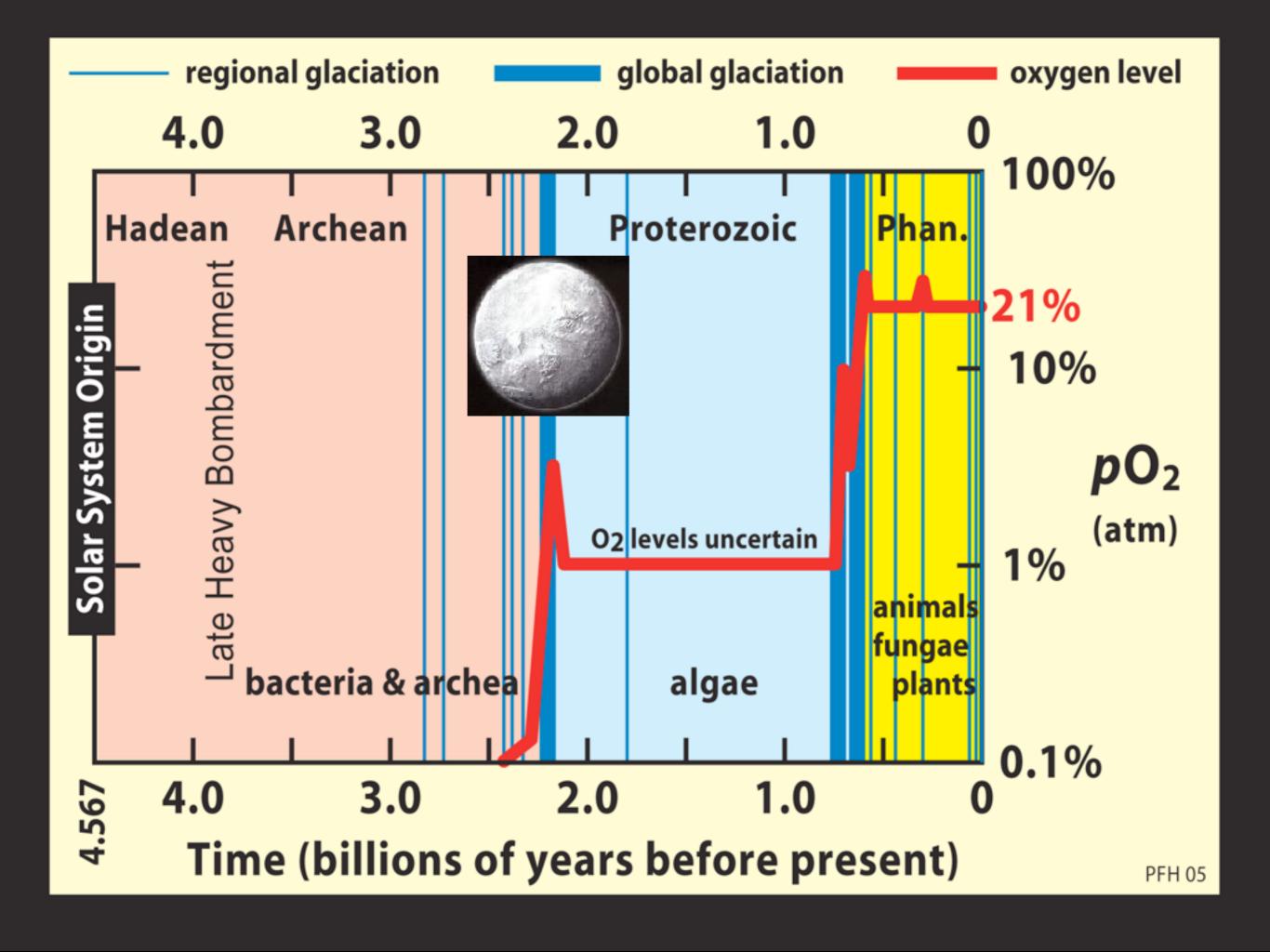
#### Carbonate-Silicate Cycle





Provides long-term climate stability

# Climate Crisis

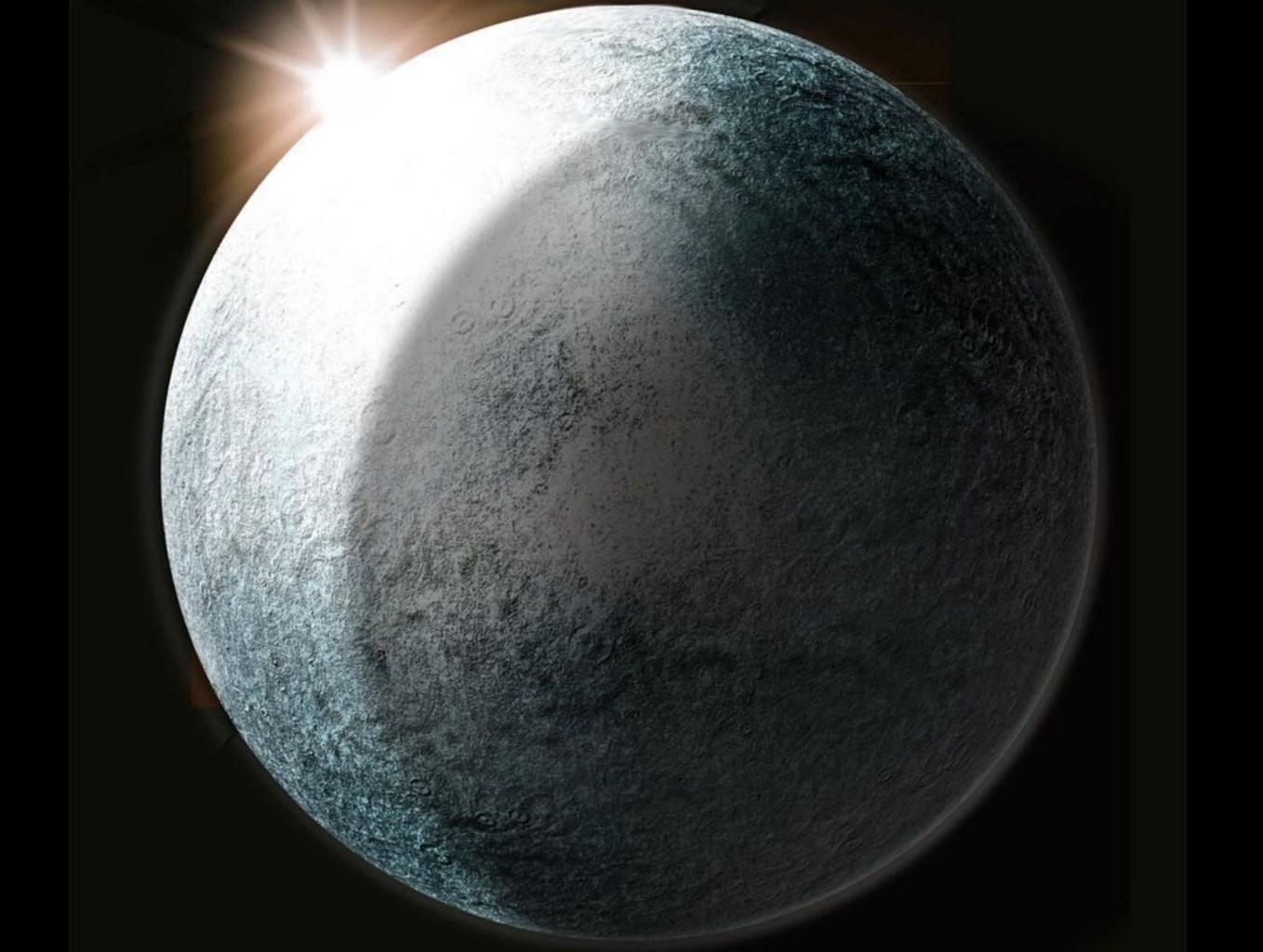


#### Snowball Earth

#### Surface Temperature

Snow and Ice cover

(+)
Planetary
Albedo



#### Geological evidence (Glaciers) for 3 total Glaciations

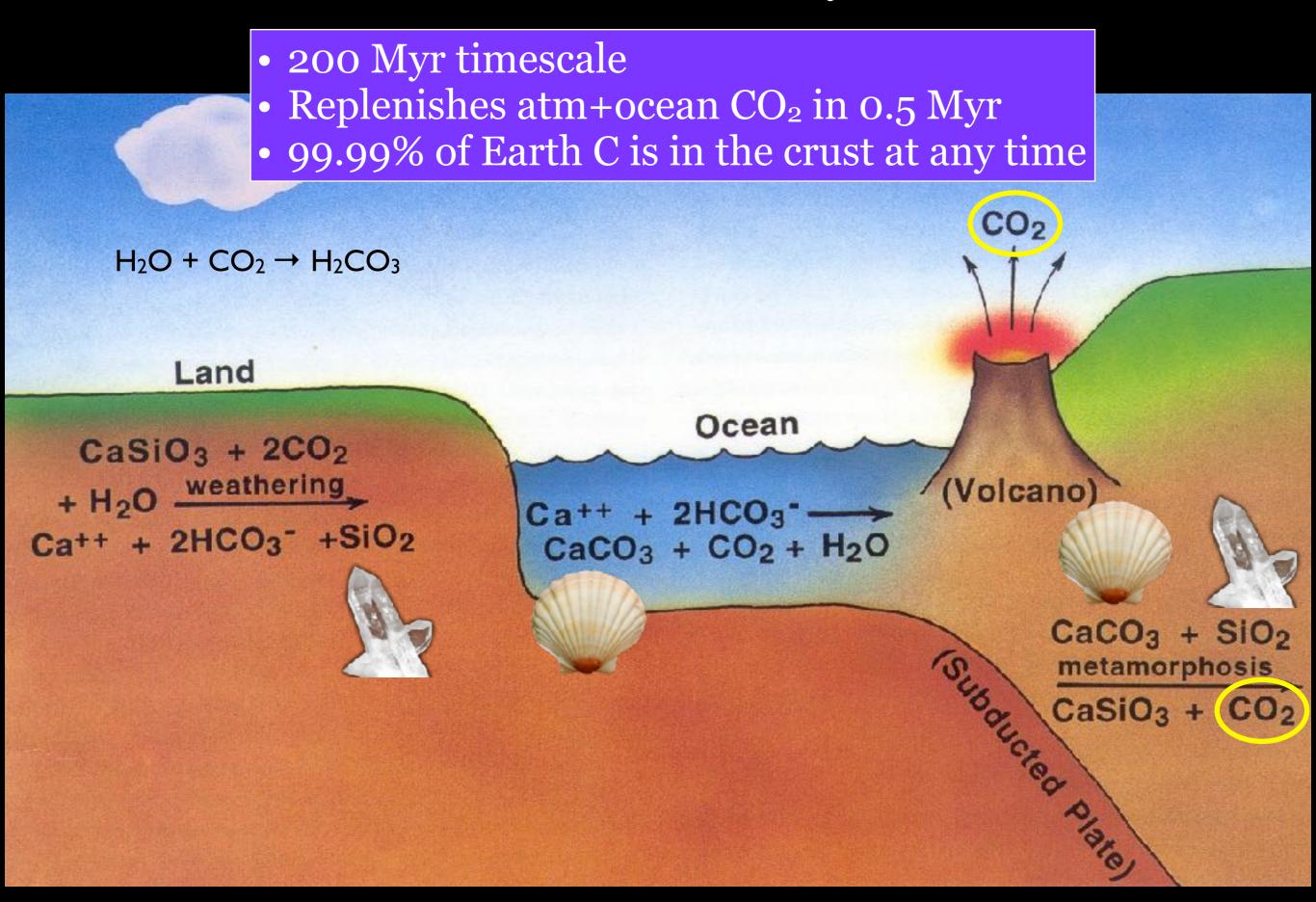
• First ~2.4-2.2 Ga (very unstable weather)

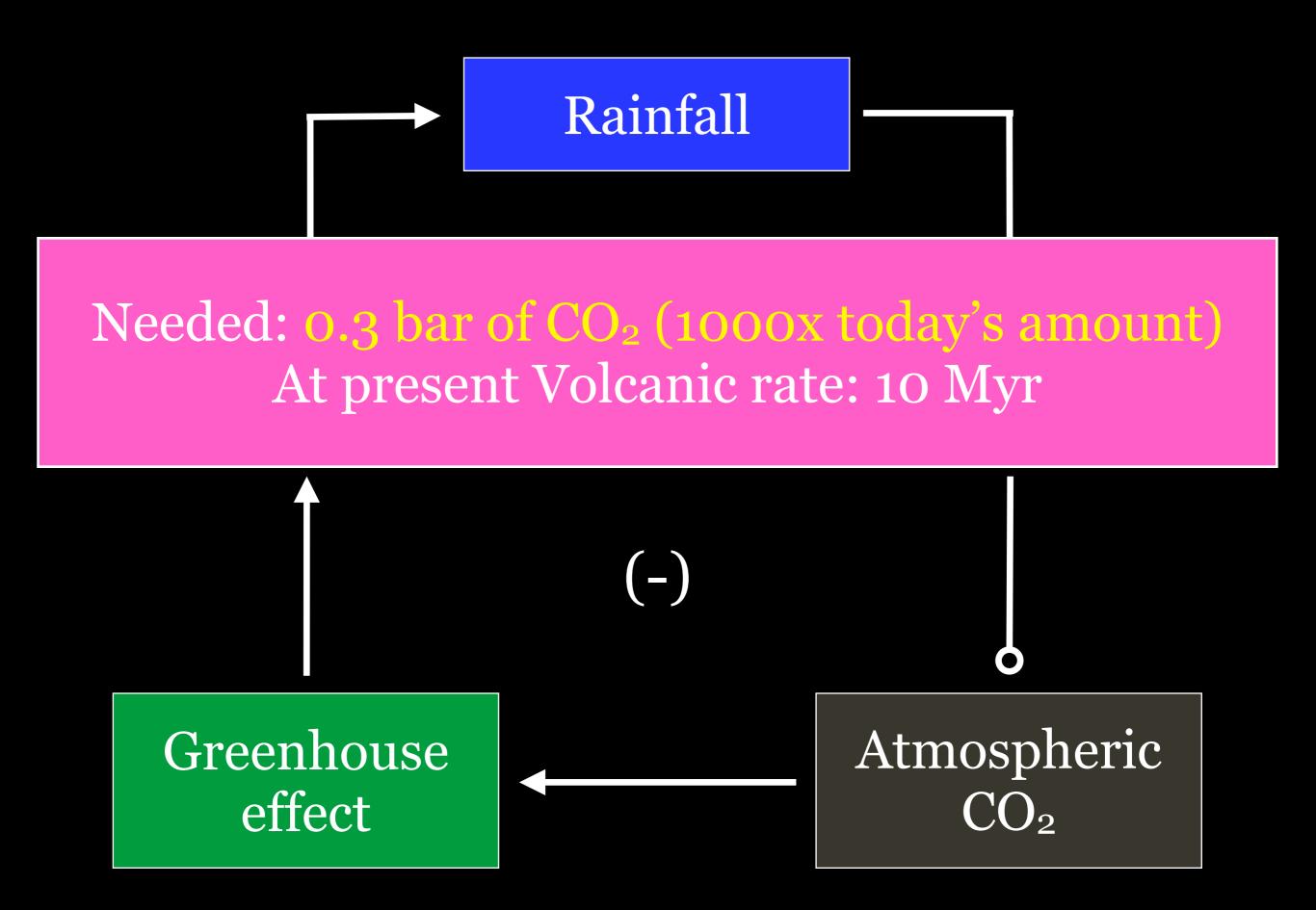
O2 ✓ → CH4 ➤ → Greenhouse effect ➤

Next 720 Ma and 580 Ma

# How did we get out of it?

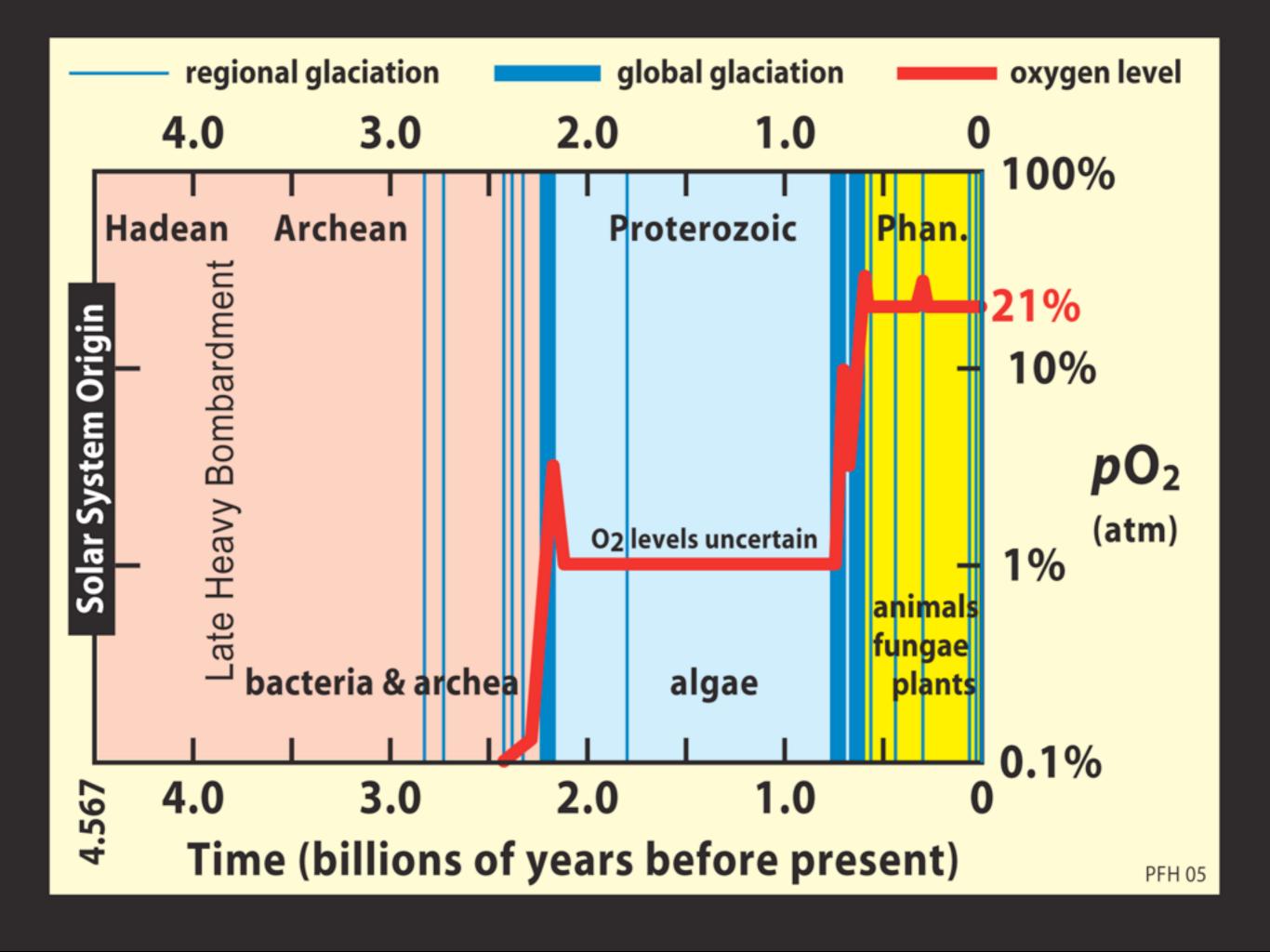
#### Carbonate-Silicate Cycle





# The Rise of Oxygen

#### Co-evolution of Oxygen and Life



### Pre-biotic Oxygen

### All the primordial atmosphere was lost during impacts

It was replaced (prior to 4.2 Ga) by a secondary atmosphere (Volcanic out-gassing, late-accretion)

Liquid water was present on the surface by then (Geological evidence)

- → Atmospheric pressure was approaching 1 bar
- → Greenhouse was possible

4.5-4.4 Ga <4.4Ga

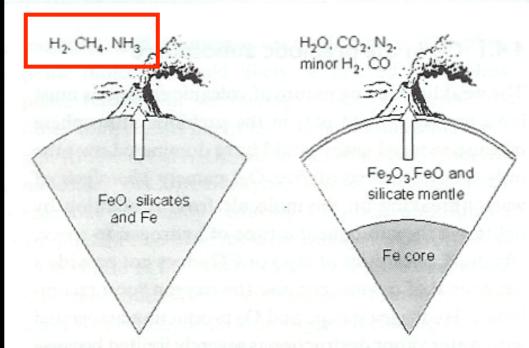


FIGURE 4.7 (Left) During 4.4—4.5 Ga when the mantle was Fe-rich during core formation, reduced volcanic gases were introduced into the atmosphere. (Right) From 4.4 Ga onwards, after the Earth had differentiated into a core, mantle, and crust, a weakly reducing mixture of volcanic gases fed the atmosphere.

Core-Mantle-Crust structure formed in the first 100 Myr

#### Where did the first abiotic O<sub>2</sub> in the atmosphere come from?

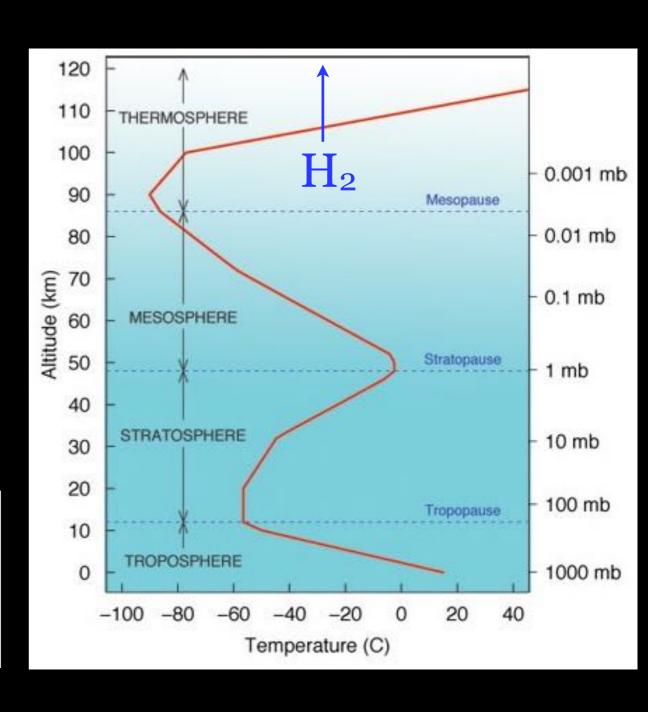
Only source: photolysis of H<sub>2</sub>O and CO<sub>2</sub>

but fast recombination

Solution: H<sub>2</sub> escapes

10-40% Thermal escape

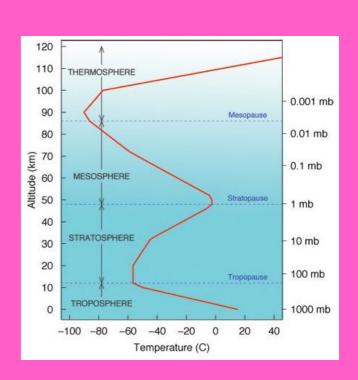
60-90% Non-thermal escape



Around ~3.9 Ga, the atmosphere reached an oxidation state close to today's value

# What was the fraction of oxygen in the Earth atmosphere prior to Life?

What did the temperature profile of the atmosphere look like?

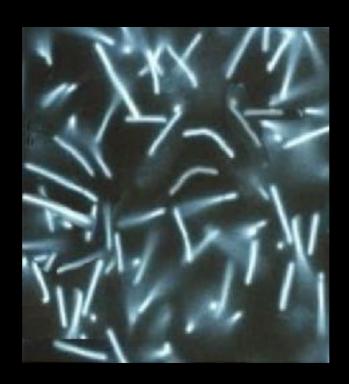


### Effects of primitive life on the atmosphere

Life modulates the most important volatiles in the atmosphere: H<sub>2</sub>, CO<sub>2</sub>, N<sub>2</sub> and O<sub>2</sub>

#### Methanogens regulate H<sub>2</sub> and CO<sub>2</sub>

$$CO_2 + 4H_2 \rightarrow CH_4 + 2H_2O$$



consume  $H_2$  and reject  $CH_4 \rightarrow Greenhouse <math>\nearrow \rightarrow$  weathering  $\nearrow \rightarrow CO_2$  in the atmosphere  $\searrow$ 

Carbon gets bound to inorganic/organic matter

## All organisms metabolize N<sub>2</sub> (If they can get it in soluble form...)

N<sub>2</sub> - Nitrogen is essential for life (DNA, RNA, proteins)

But: very few organisms can metabolize N<sub>2</sub> directly from the atmosphere

Solution:  $N_2 + CO_2 \rightarrow 2NO + 2CO$ (with the help of lightning in the atmosphere)

NO is soluble in water (HNO), can be fixed (anaerobic) and then be used by organisms

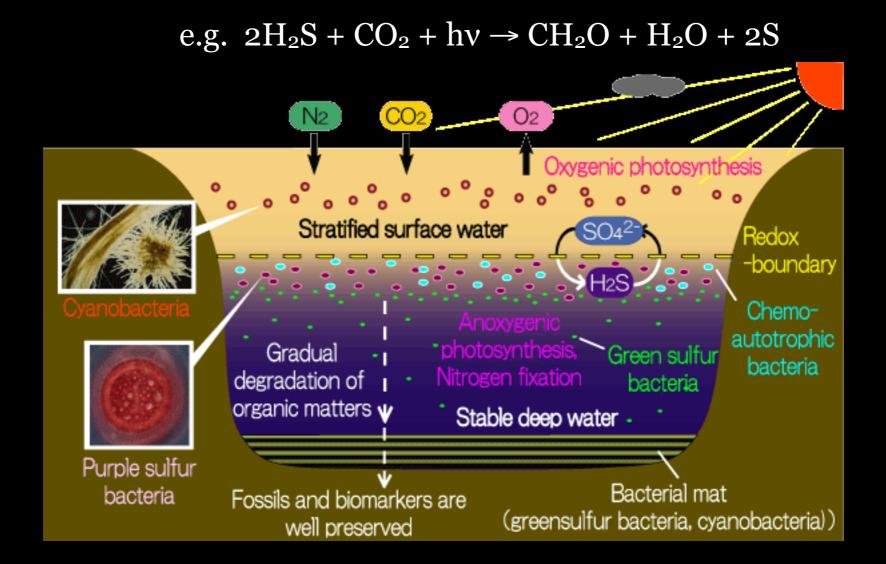


### Take a break...

### The first Rise of Oxygen

#### Anoxygenic Photosynthesis

- pre-dates Oxygen photosynthesis (probably 3.5-3.2 Ga)
- principle: use sunlight to extract protons (H+) from H<sub>2</sub>S, store energy in ATP, and use it to extract C by reducing CO<sub>2</sub>

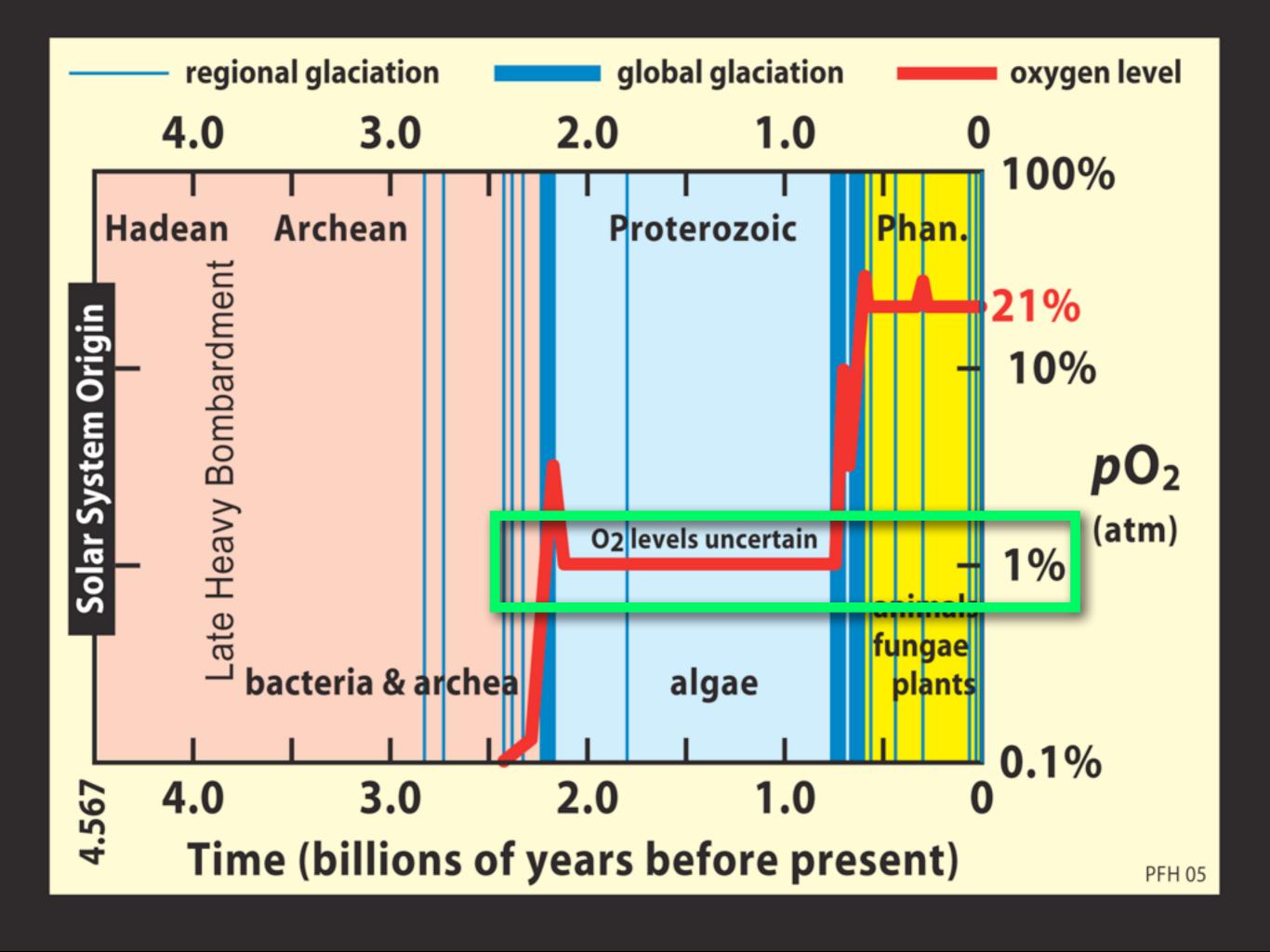


There are no pockets of "old" atmosphere left. How can we trace the early atmosphere?

How old are the oldest ice cores?

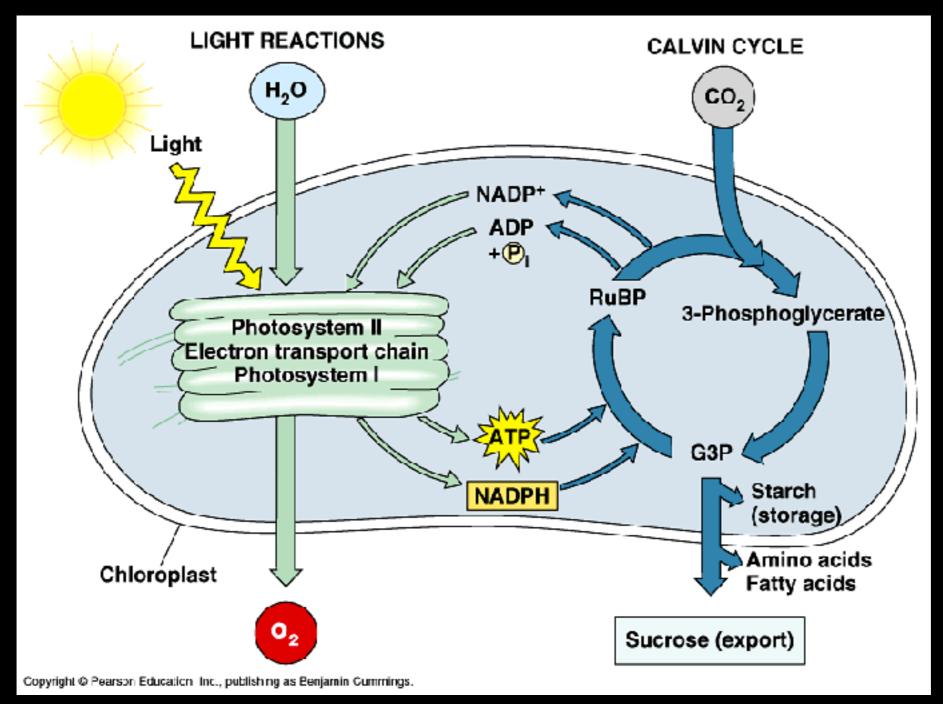
### Geological evidence for O2 in atmosphere





#### Oxygenic Photosynthesis

- probably invented ~2.7 Ga (evidence in Stromatolites)
- principle: use sunlight to extract protons (H+) from H<sub>2</sub>O, store energy in ATP, and use it to extract C by reducing CO<sub>2</sub>

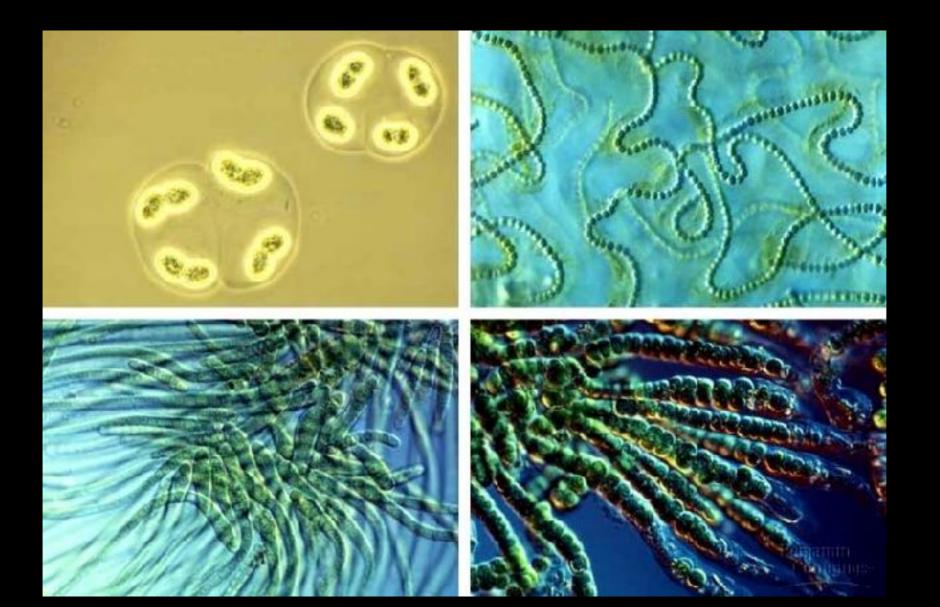


First to use it: Cyanobacteria

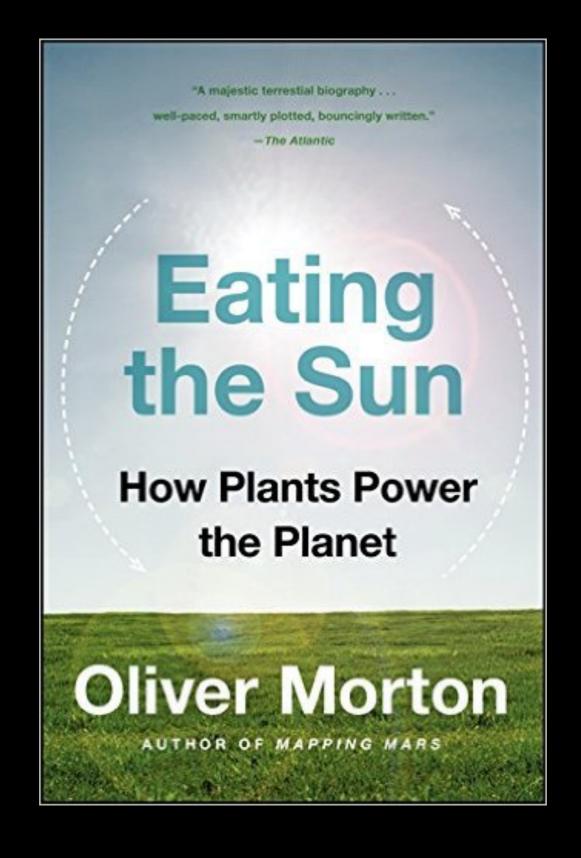
Found 0.3-0.4 Gyr before the rise of Oxygen!

Main trick needed: overcome toxic O<sub>2</sub>

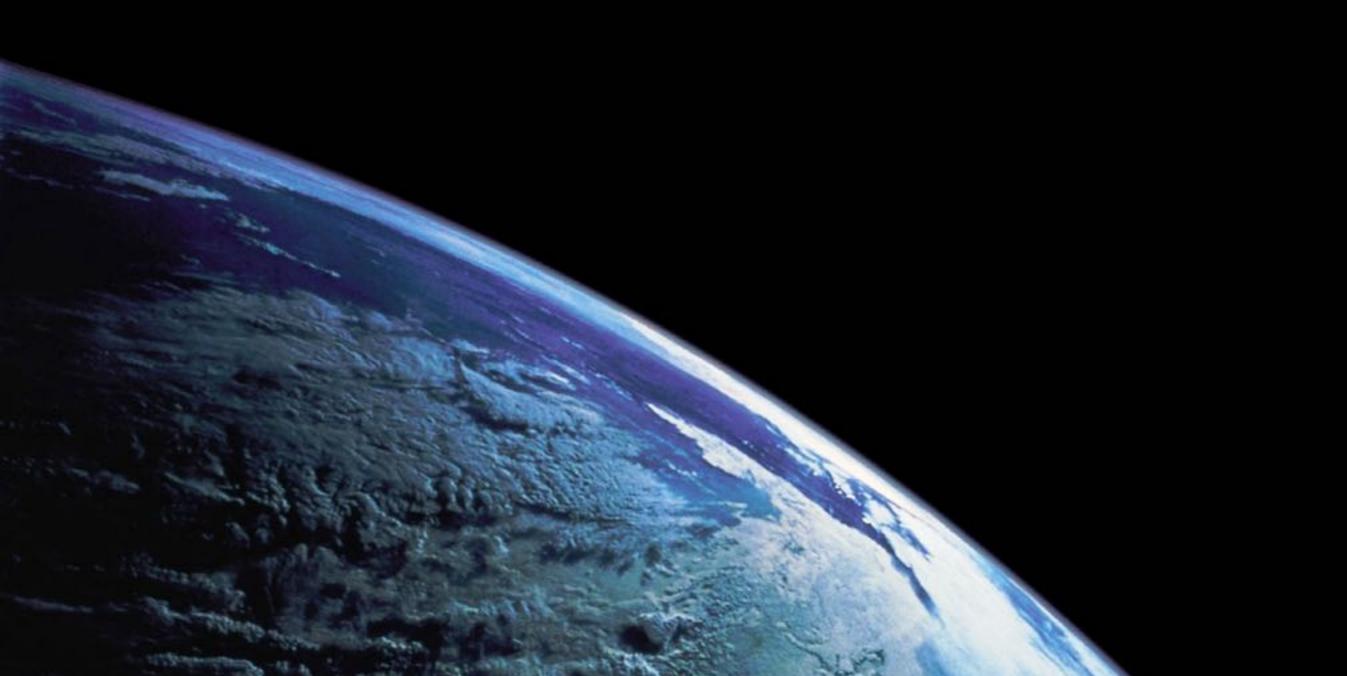
Today found as Chloroplast in Eukaryotes



### Today found as Chloroplast in Eukaryotes

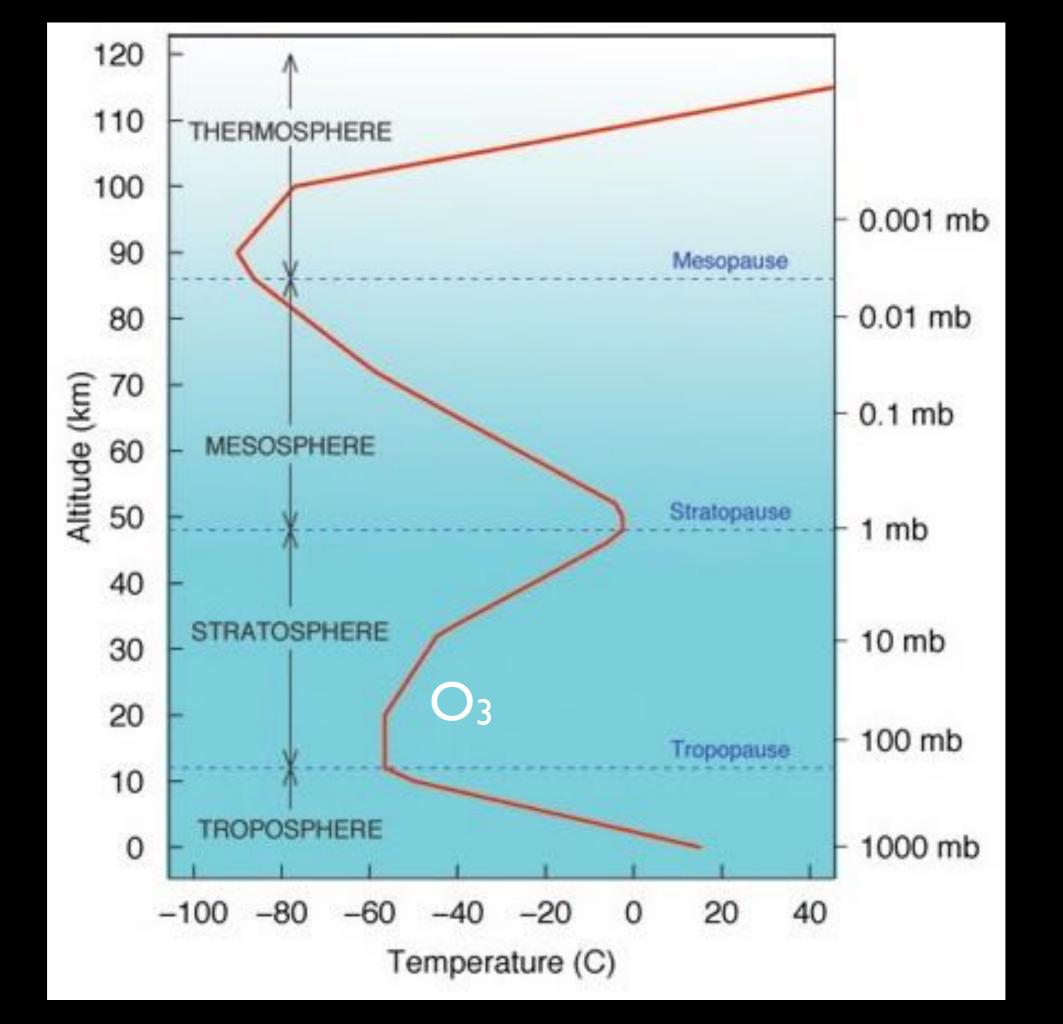


### With Oxygen comes Ozone...



# What is the effect of Ozone on the stratosphere?

## Why is Ozone important for Life?

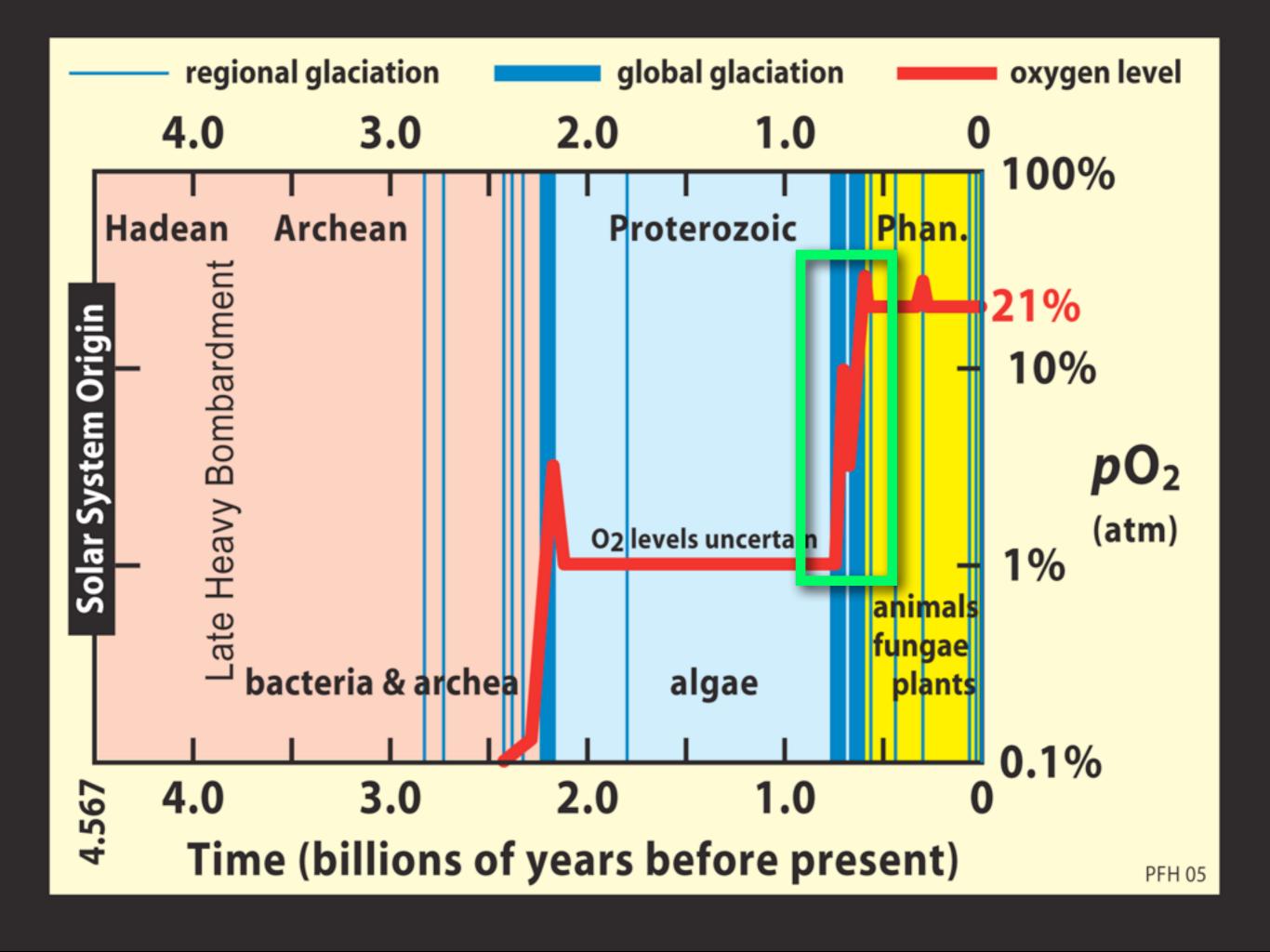


### The Ozone shield → allows life at the surface

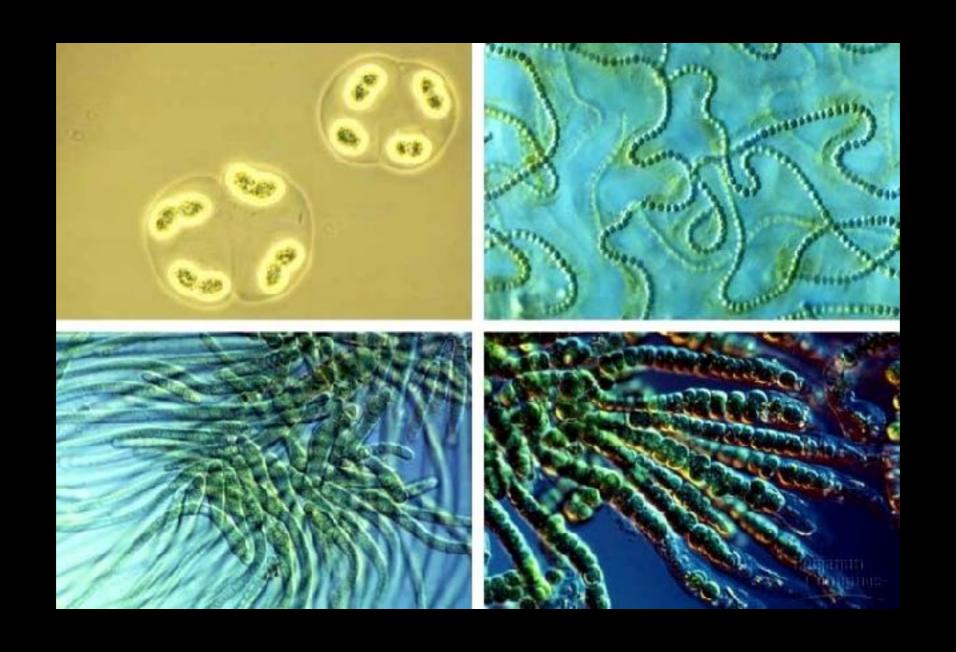
The Ozon layer appeared with the rise of Oxygen (~2.4-2.3 Ga ago) in the lower Stratosphere (10-20km)

 $CO_2$  protects < 200nm  $O_3$  protects 200-300 nm (already at 1% of todays  $O_2$ )

### The next Rise of Oxygen

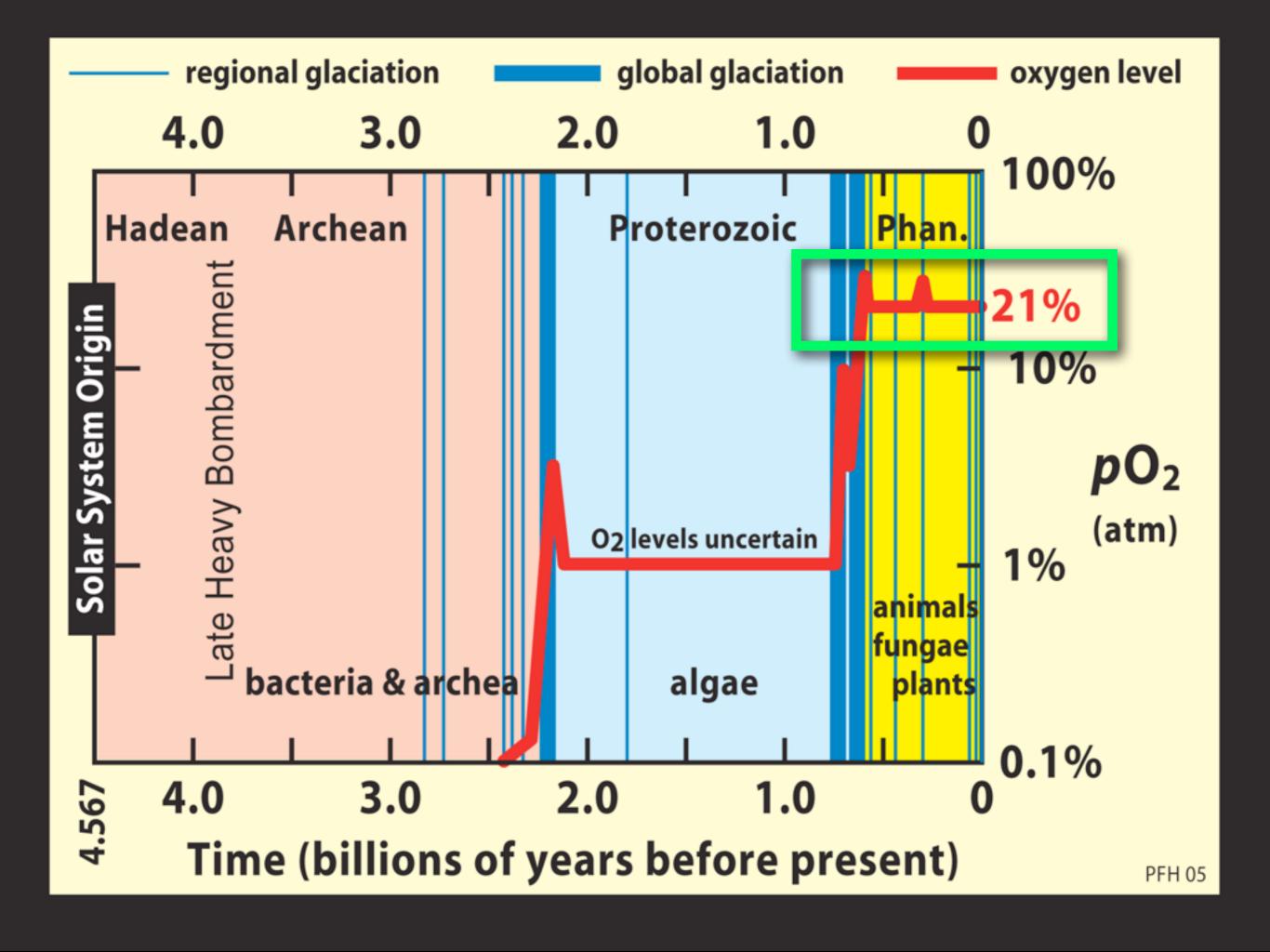


## Most likely, organism with Oxygenic Photosynthesis expanded rapidly after a (near) total glaciation...





# Who are the main producers of oxygen today?



### When is a Planet "habitable"?



# When would you claim that a planet is habitable?

### Water is important

- For all biological processes
- For stability of the climate
- For short-term temperature stability (heat capacity)



In order to detect LIFE in the atmosphere it is a lot easier if it is on the surface of the Planet



Start the search where Water is liquid on the surface

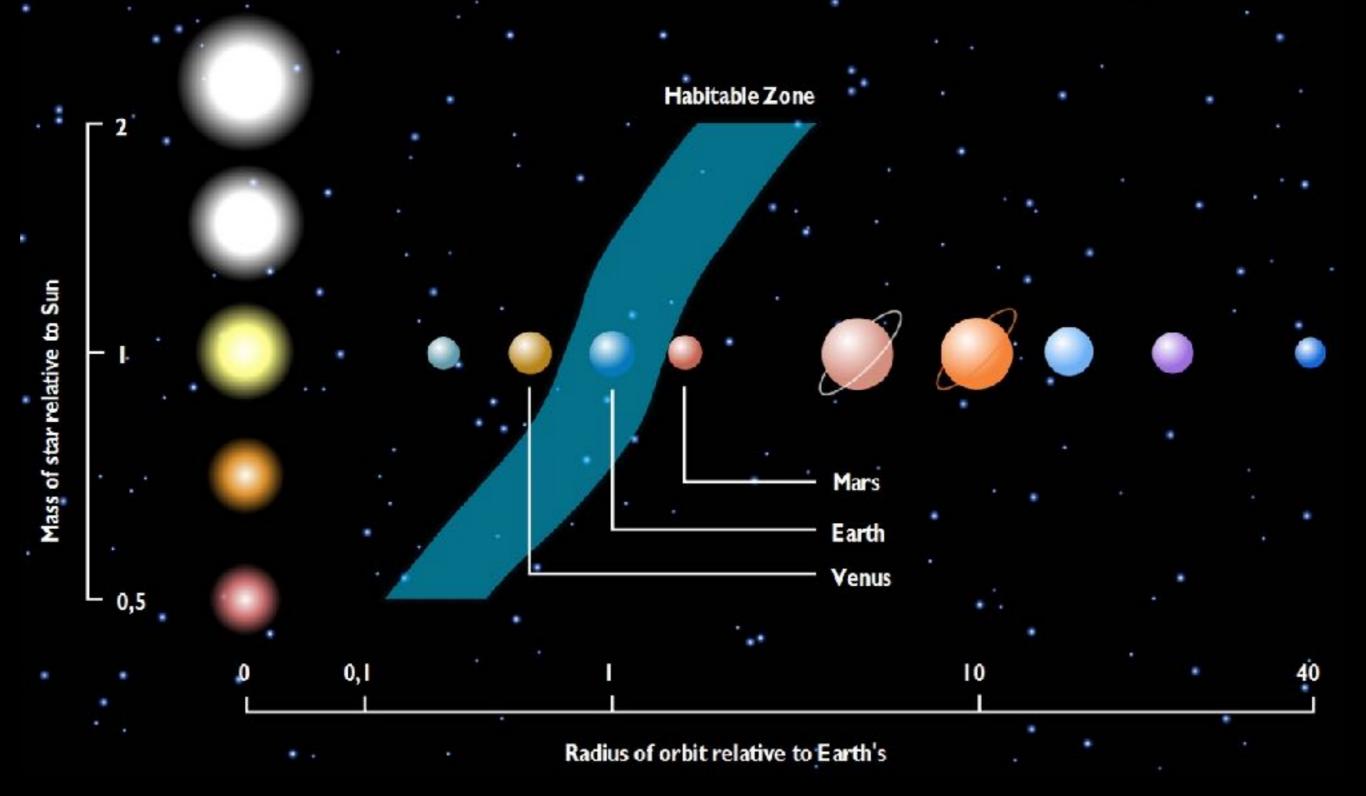
A planet is habitable if it provides the environment, materials and processes that are advantageous for the formation and long-term evolution of life

The Habitable zone was first know as "Liquid Water Belt" (Shapley 1953), "Ecosphere around the Sun" (Strughold 1953), before

"Habitable zone" (Su-Shu Huang 1959)

Often Human-centric view for life as we know it

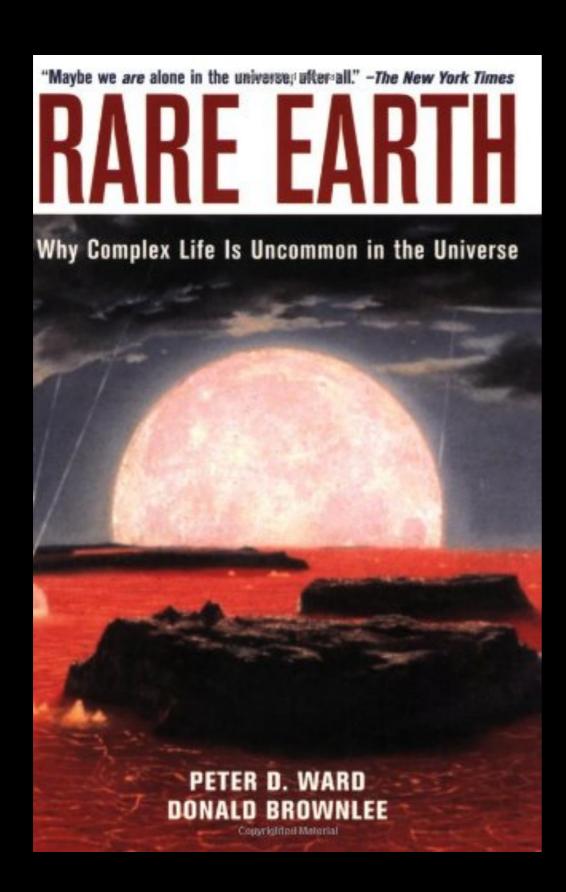
Habitability is more than "liquid water on surface"

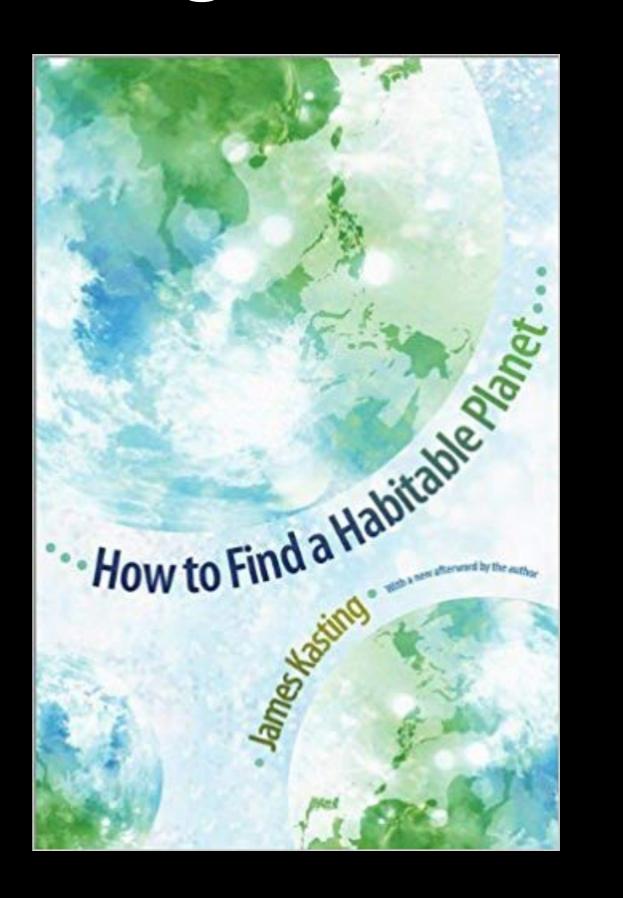


# Why should we search on Venus and Mars, when they are outside the "habitable zone"?

# Other Factors that influence Habitability...

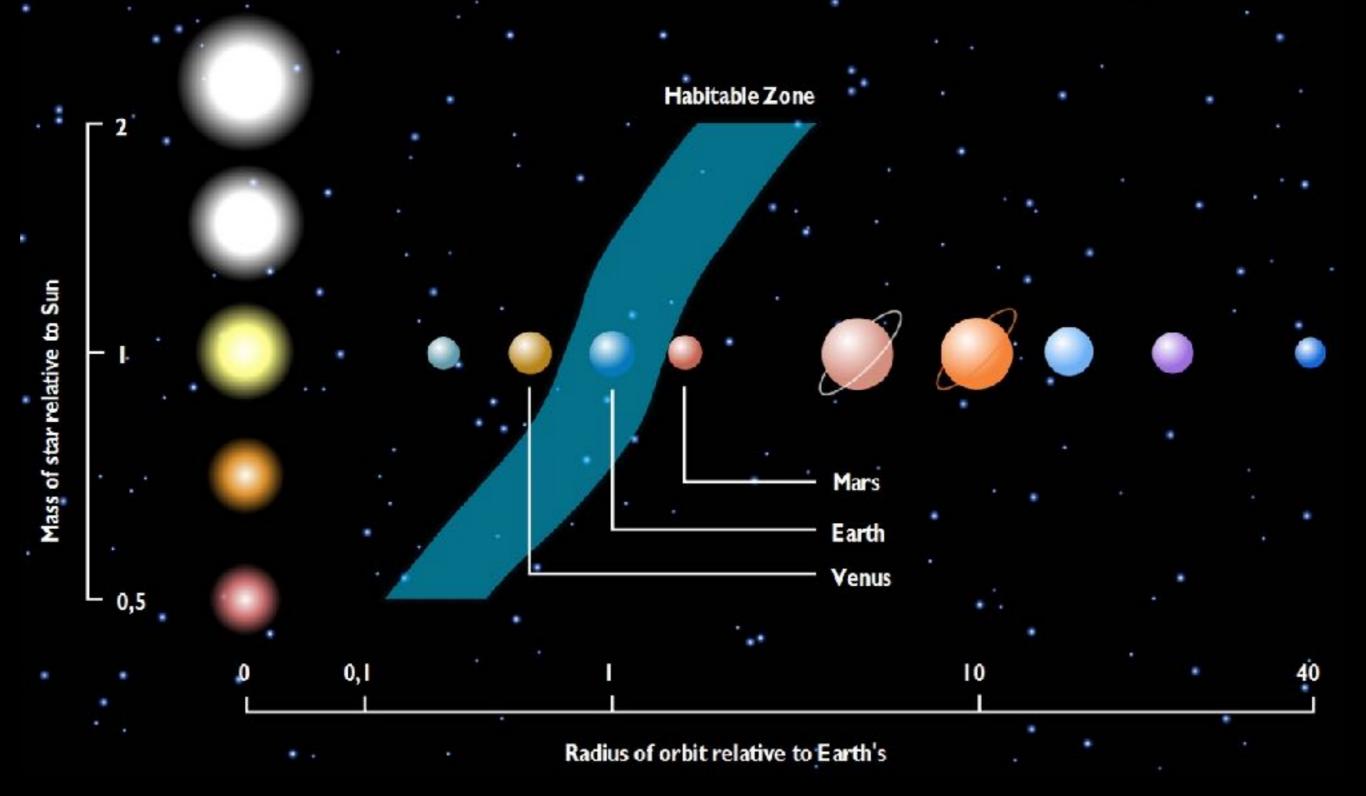
### Re-considering...





### Lunch break...

## Runaway Greenhouse: The Evolution of Venus



#### Venus: Earth "Sister"

Distance from Sun: 0.72 AU

Mass: 81% of Earth

#### But....

 $T_{surf} = 460 \, ^{o}C$  (too hot for liquid water, even under high pressure)

P<sub>surf</sub> ~ 93 bars

ATM: CO<sub>2</sub> (96.5%), N<sub>2</sub> (3.5%), traces of SO<sub>2</sub>, H<sub>2</sub>O, CO leading to H<sub>2</sub>SO<sub>4</sub> rain

Secure inner limit of the habitable zone

UV image

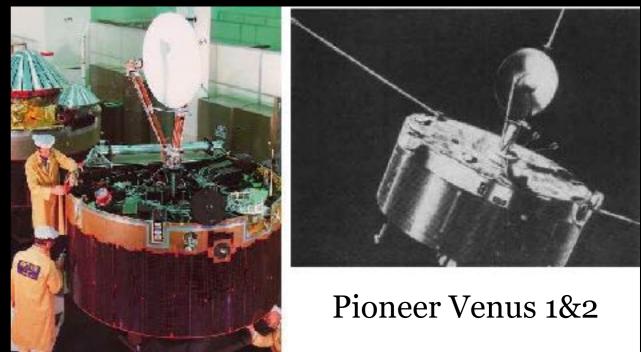
#### Did Venus start wet or dry?

#### Pioneer Venus (1977/1978):

today: only 10-3 x Earth H<sub>2</sub>O (in atm 30 ppm vs. 1000-40,000 ppm)

but D/H ratio: 150 x Earth

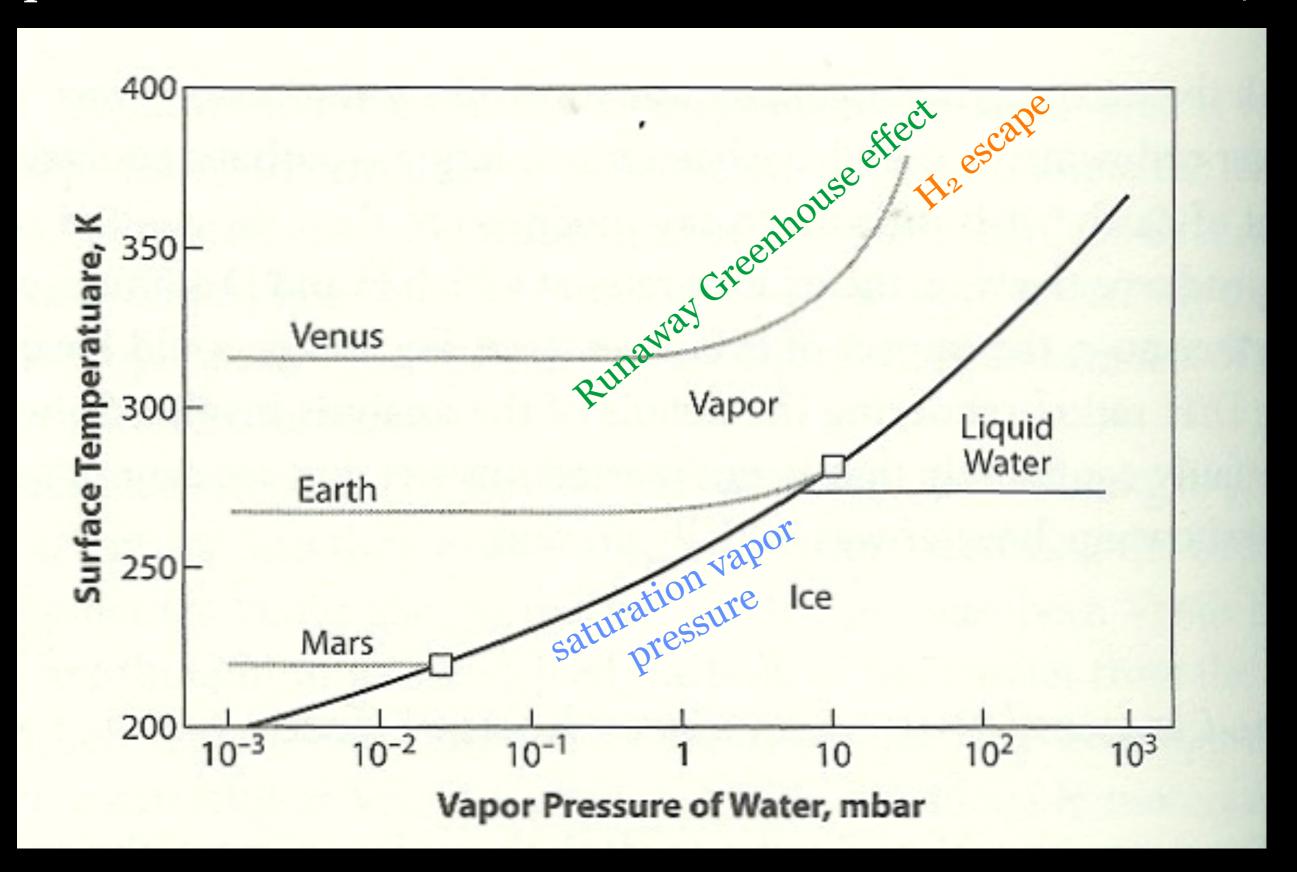
H<sup>+</sup> escapes, D<sup>+</sup> (heavier) less...



Venus started with surface water!

How much is unclear...

# A simple model based on no initial atmosphere, pure $H_2O$ volcanism, and all albedos = Mars' albedo = 0.17



# How much more radiation flux from the Sun does Venus get (at 0.72 AU) than Earth?

#### Gedankenexperiment: moving Earth closer to the sun

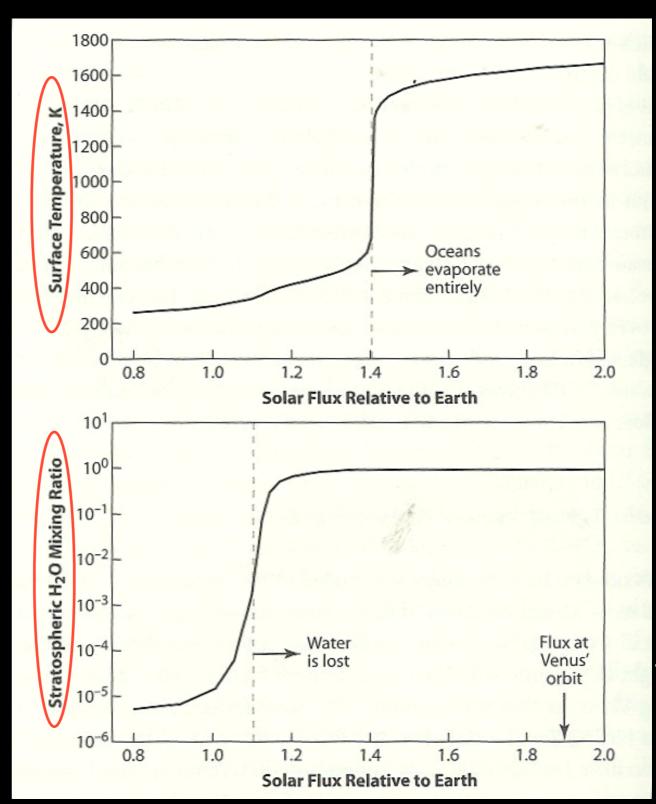
At ~1.4 x Earth solar flux (i.e. 0.85 AU):

Runaway Greenhouse Oceans evaporate entirely

At ~1.1 x Earth solar flux (i.e. 0.95 AU):

The tropopause is lifted (from 10-15km to 150km) A wet stratosphere develops and H<sub>2</sub>O is lost

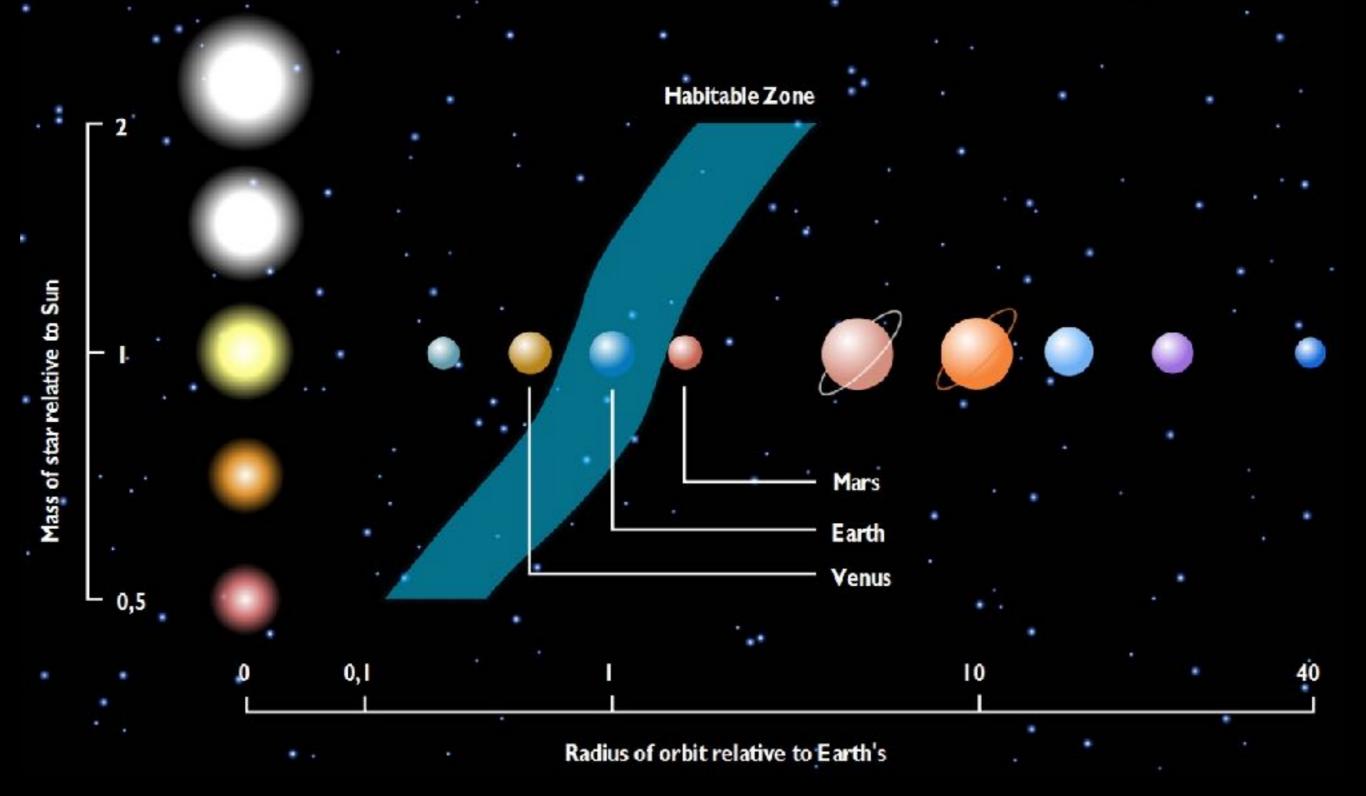
This is the inner edge of the Habitable Zone



cold trap is lost Ozone is destroyed

# Climate History of Mars

# How did the inner/outer edge vary as a function of time?



#### Mars: the red planet

Distance from Sun: 1.52 AU

Mass: 11% of Earth

#### and....

 $T_{surf} = -55$  °C  $P_{surf} \sim 6-8$  mbar (water sublimates)

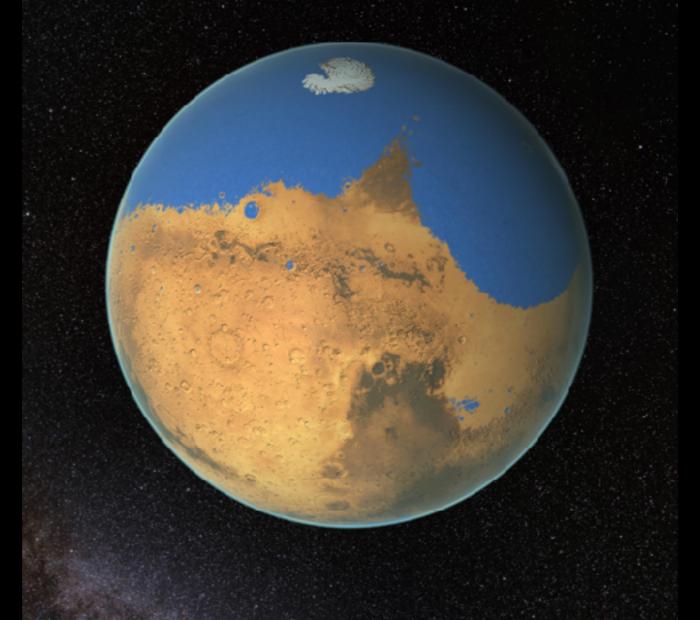
ATM: CO<sub>2</sub> (95.3%), N<sub>2</sub> (2.7%), Ar (1.6%) traces of O<sub>2</sub>, CO, H<sub>2</sub>O

Obliquity changes by  $\pm$  10° in 10<sup>5</sup> -10<sup>6</sup> yr cycles chaotically and can range from 0° to 60°

#### Did it ever rain on Mars?

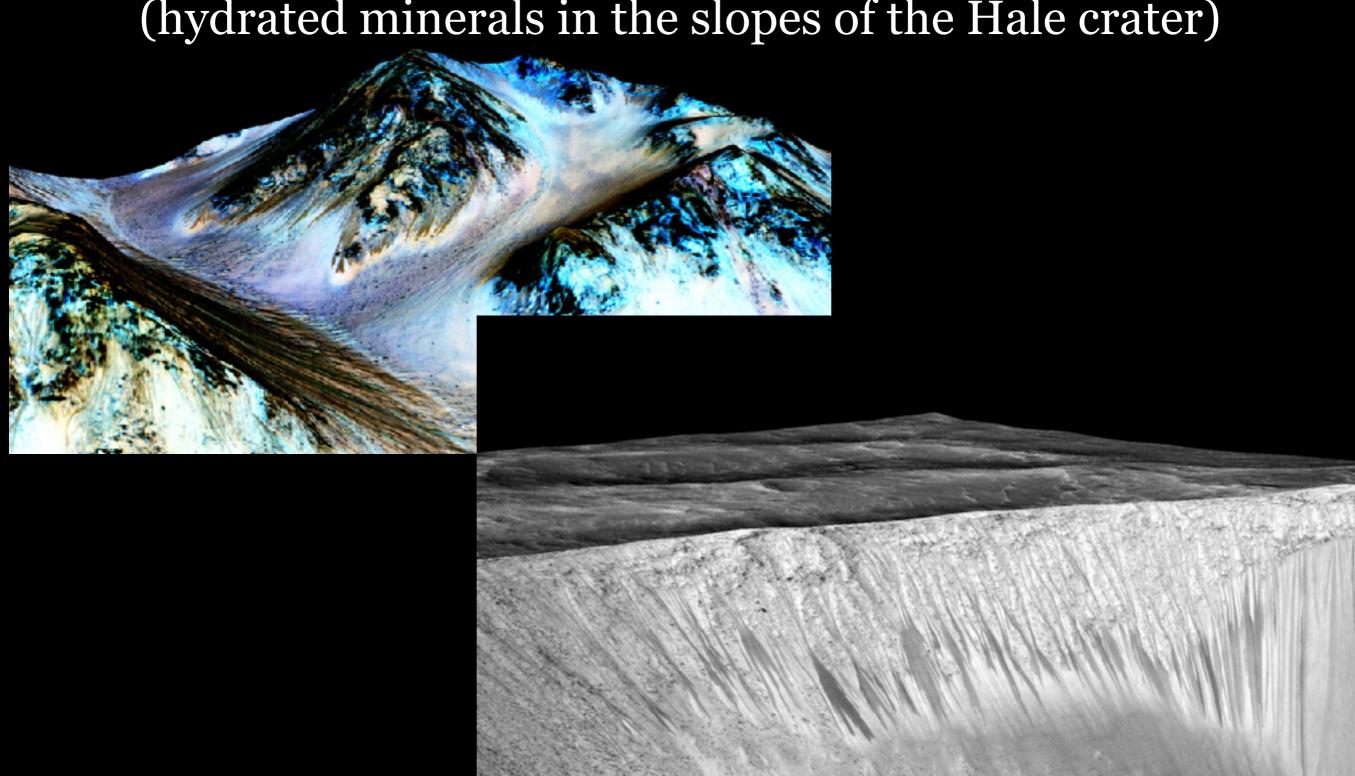
Comparison of HDO with H<sub>2</sub>O in water on Mars today vs. Mars meteorites dating from 4.5 Ga ago

Deduced that 20 million km<sup>3</sup> of water were liquid (1/70 of the amount on Earth today), covering 20% of Mars



#### Did it ever rain on Mars?

The Mars Reconnaissance Orbiter, in 2015, confirmed evidence that water flows on Mars today (hydrated minerals in the slopes of the Hale crater)



#### Could early Mars have been warm?

Earth analogy: early Greenhouse gases were CO<sub>2</sub> and CH<sub>4</sub>

CO<sub>2</sub>: through Volcanism 🗸

CH<sub>4</sub>: abiotic? biotic (methanogens)? ✓

SO<sub>2</sub>: Volcanic origin but unstable in the atmosphere (Note: leads to H<sub>2</sub>SO<sub>3</sub> and acid surface, explains absence of carbonates)

#### but...

at <3.8 Ga and 1.52 AU, Solar intensity was 32% of Earth today

 $\rightarrow$  Max.  $T_{surf} \sim 225 K (-48 °C)$ 

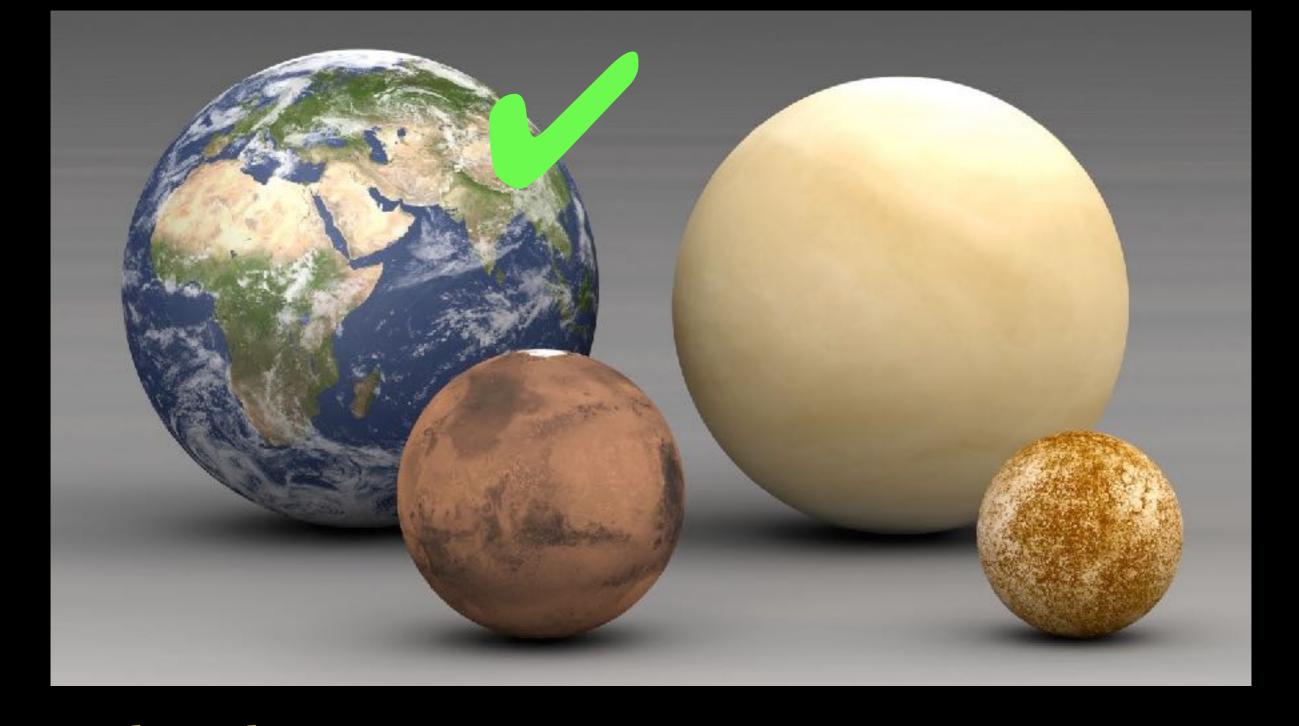
#### Today Volcanism stopped on Mars



If this were not the case (if Mars were more massive and had tectonic activity) there would be enough CO<sub>2</sub>

At ~0.7 x Earth solar flux (i.e. 1.2 AU): CO<sub>2</sub> condenses to clouds and the greenhouse effect is reduced

The outer edge of the Habitable Zone is fuzzy but around 1.5 AU



In the Solar System:

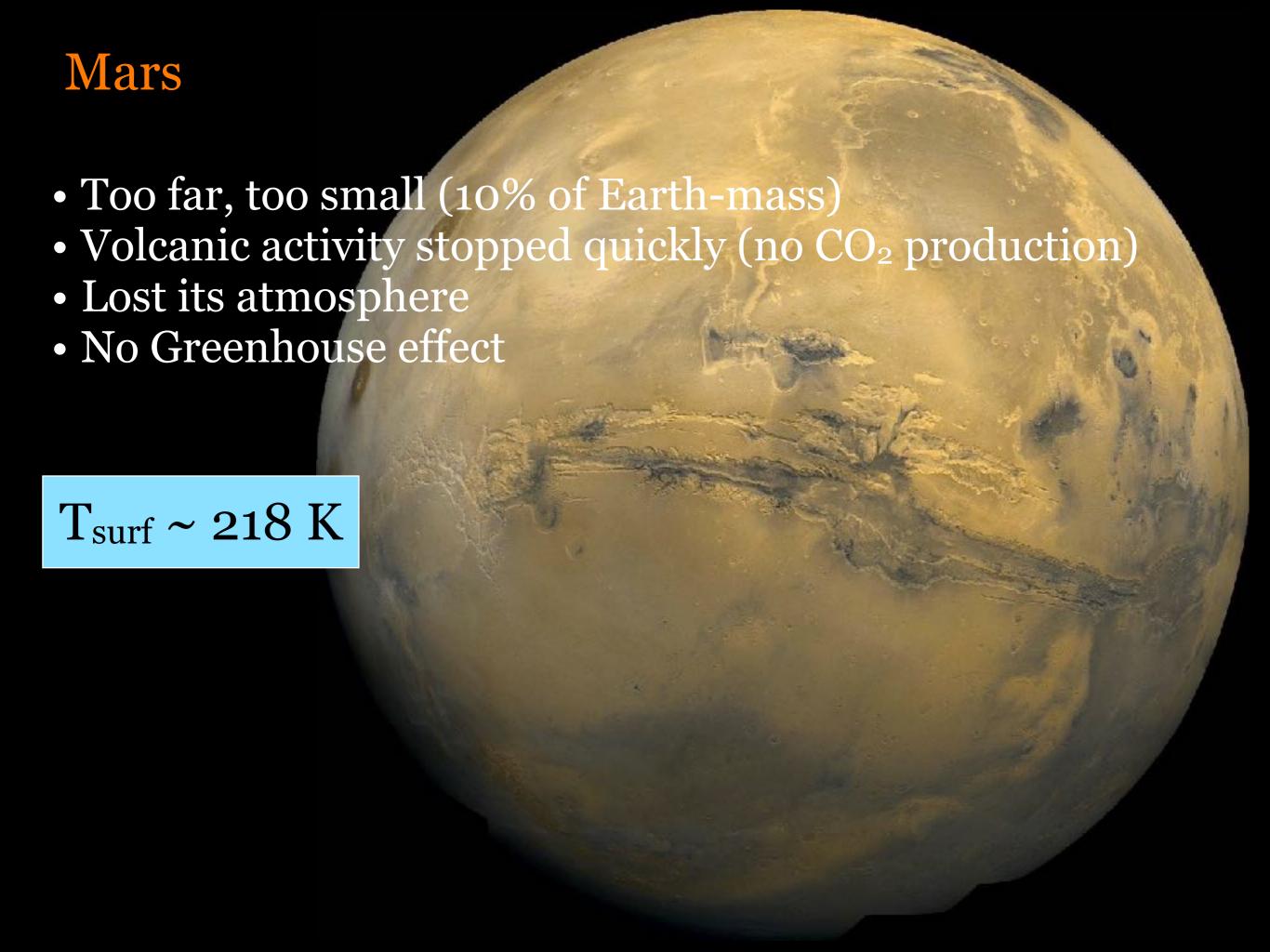
The inner edge of the Habitable Zone is around ~0.9 AU

The outer edge is fuzzy but around ~1.5 AU

#### Venus

- Close to Sun
- Lost its H<sub>2</sub>O
- No weathering (CO<sub>2</sub> stay in atm)
- Runaway Greenhouse effect

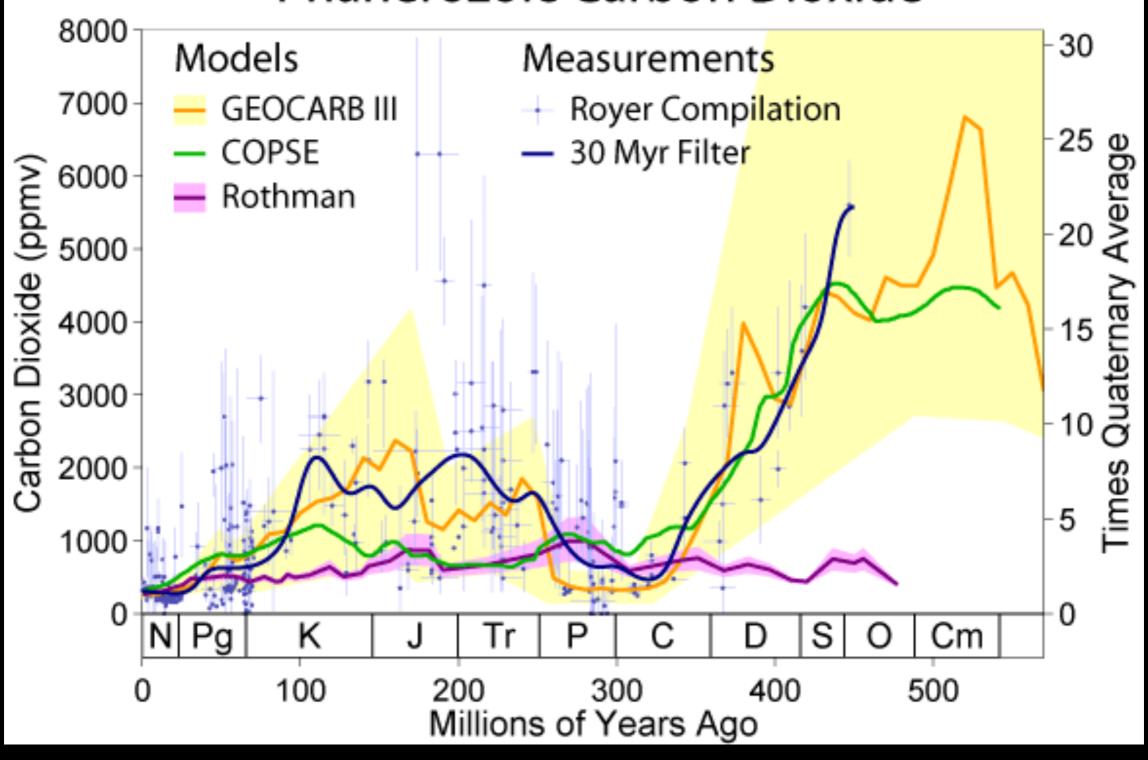
 $T_{surf} \sim 730K$ 



# What about the climate on Mercury?

### The Future of Earth

#### Phanerozoic Carbon Dioxide



# What will be the long-term consequence on the climate of today's raise in CO<sub>2</sub>?

And the consequence for Life on Earth?

Current:  $CO_2$  ~400 ppm ( $CO_2$  is an efficient Rayleigh scattered = higher Albedo)

If we burned all fossil reserves: 1400-2000 ppm



- → higher Greenhouse (compensated partly by higher albedo: no runaway process)
- $\rightarrow$  T<sub>surf</sub> + ~8°C (sea level + ~8om)

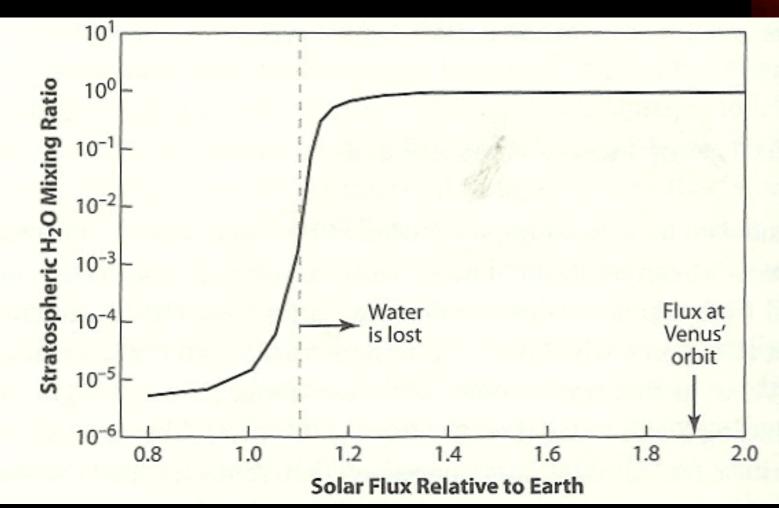
Even at Venus CO₂ concentration: T<sub>surf</sub> ~230°C but
Pressure ✓ (100 bar) → Water remains liquid

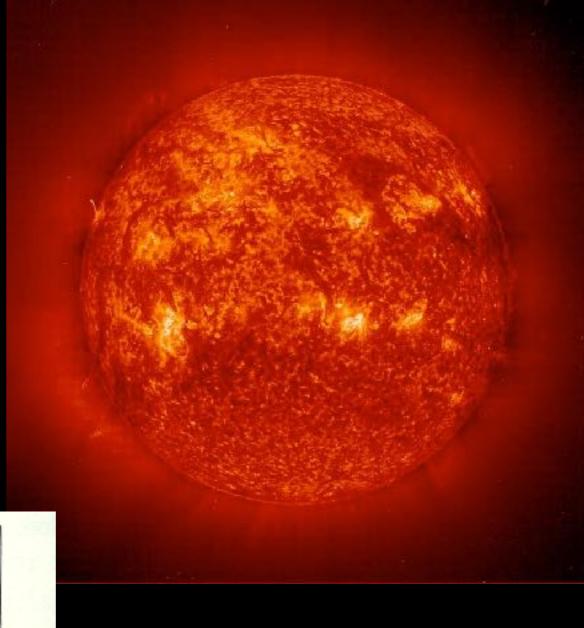
Today, Earth cannot lose its water (Humans would nevertheless feel very uncomfortable)

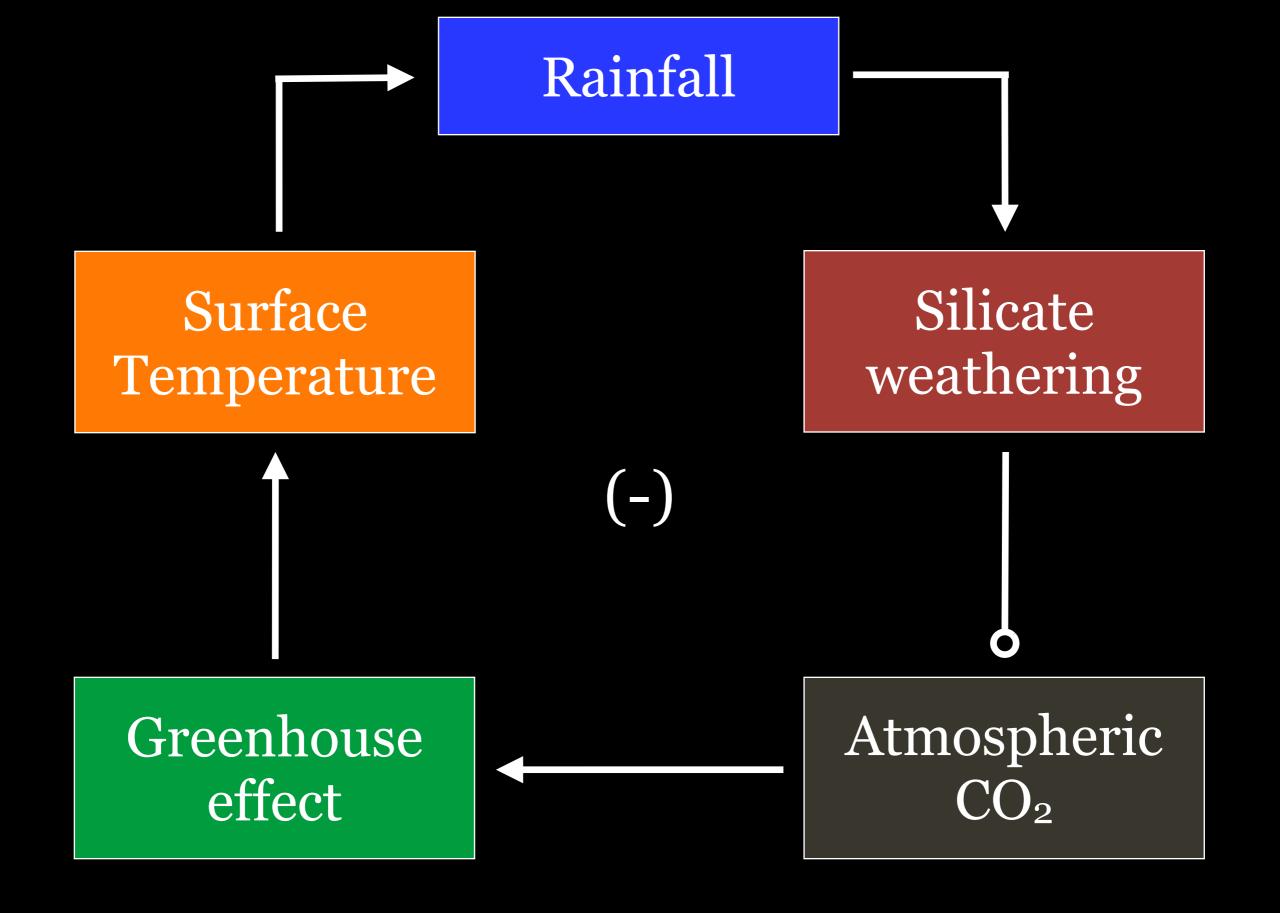
#### But what about in ~1 Gyr time?

Solar luminosity up by 10%

⇒ Earth loses its water but how fast?







Provides long-term climate stability

#### What will happen?

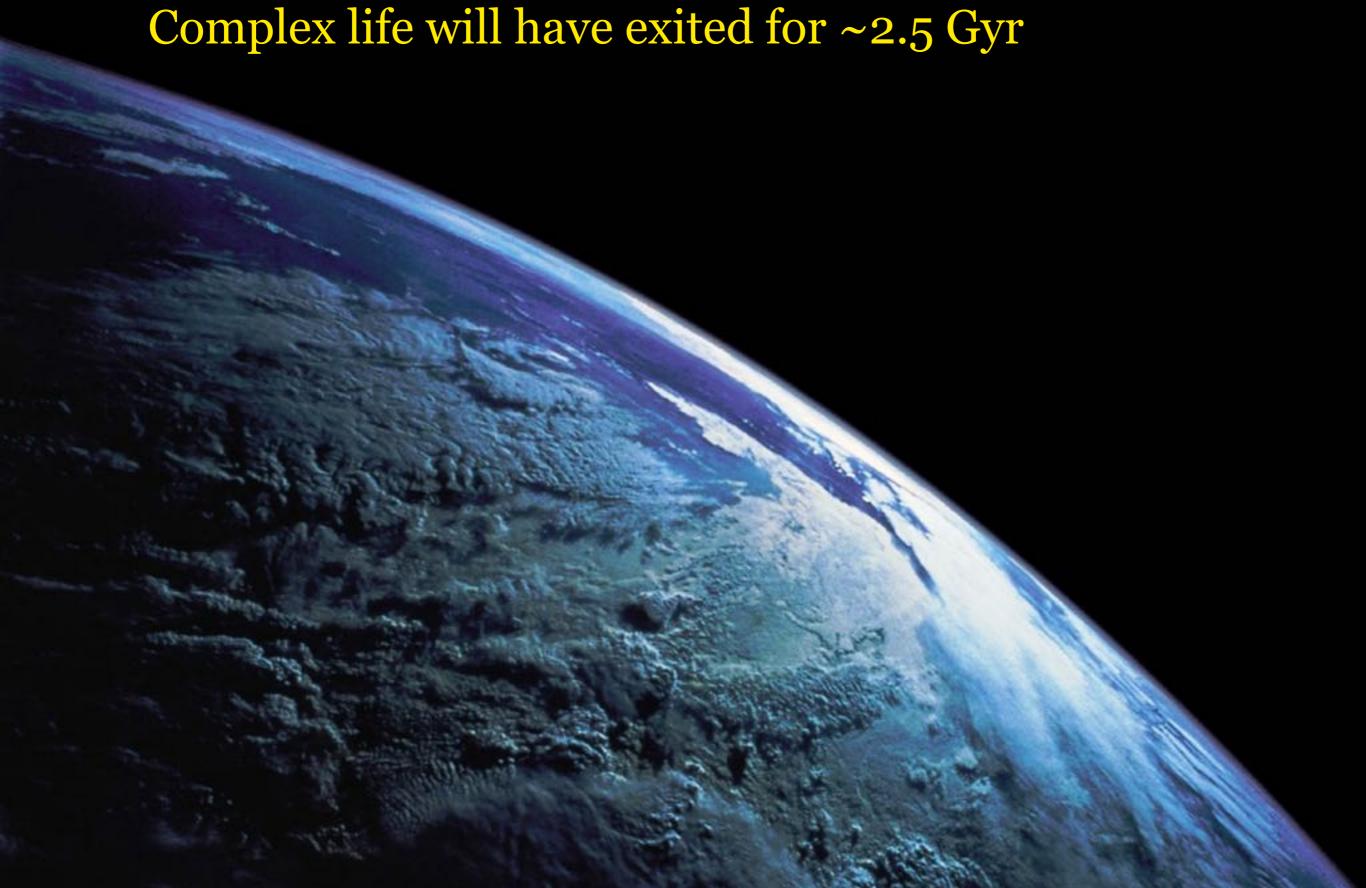
(ignore the short term human-made CO<sub>2</sub>)

Solar luminosity  $\nearrow$   $\rightarrow$  weathering increases  $\rightarrow$  CO<sub>2</sub>  $\searrow$ 

- at 150 ppm CO<sub>2</sub> (after 500 Myr) C3 plants die C3 plants: all trees, most crops [95%]
- at 10 ppm CO<sub>2</sub> (after 900 Myr) C4 plants die C4 plants: tropical plants, corn, sugar cane [5%]

The atmosphere becomes thin (low pressure) and H<sub>2</sub>O gets lost over a few 100 Myr

Earth lifetime with surface H<sub>2</sub>O will have been ~5.5 Gyr





#### Take home ideas from day 2

- Solar flux, albedo, greenhouse effect
- Feedback loops drive the climate
- Life modulates the most important volatiles in the atmosphere
- Oxygen appeared with photosynthesis
- Habitability is (currently) linked to liquid water
- Venus experienced a runaway greenhouse effect
- Mars never developed a greenhouse atm

#### Homework:

- Watch the BBC documentary 'Earth The Power of the Planet', episode 2 'Atmosphere' (<a href="http://documentaryheaven.com/atmosphere-earth-the-power-of-the-planet/">http://documentaryheaven.com/atmosphere-earth-the-power-of-the-planet/</a>)
  If motivated, look at episodes 1 and 5: 'Volcano' and 'Rare Earth'
- Read one (or both) Chapters of 'The Emerald Planet' by D.Beerling Start with 'Oxygen and the lost world of giants', and if motivated, continue with 'Nature's Green Revolution'

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Friday November 24	Day 5: Search for Extraterrestrial Intelligence; Alien Biochemistry 10:00-12:00 & 13:00-14:00

# The End for Today

# Thank you!